

Module 1: Chromatography

5th Grade

About the Instructions:

This document is intended for use by classroom teachers, SciTrek Leads, and SciTrek Volunteers. The document has been composed with input from teachers, leads, volunteers, and SciTrek staff to provide suggestions to future teachers/leads/volunteers. The instructions are not intended to be used as a direct script but were written to provide teachers/leads/volunteers with a guide line to present the information that has worked in the past. Teachers/leads/volunteers should feel free to deviate from the instructions to help students reach the learning objectives of the module. Some places in which you can be creative and mold the program to meet your individual teaching style are: during class discussion, group management, generating alternative examples, and asking students leading questions. However, while running the module make sure to cover all the material each day within the given 60 minutes. In addition, no changes should be made to the academic language surrounding conclusions or the conclusion activity.

Activity Schedule: There are no scheduling restrictions for this activity.

Day 1: Conclusion Assessment/Observations/Variables (60 minutes)

Day 2: Question/Experimental Set-Up/Procedure (60 minutes)

Day 3: Experiment/Conclusion Activity (60 minutes)

Day 4: Conclusion Activity/Conclusion/Question/Experimental Set-Up/Procedure (60 minutes)

Day 5: Experiment/Conclusion (60 minutes)

Day 6: Peer Conclusion Check/Poster Making (60 minutes)

Day 7: Poster Presentations (60 minutes)

Day 8: Conclusion Assessment/Tie to Standards (60 minutes)

NGSS Standard Addressed:

5-PS1-3 Make observations and measurements to identify materials based on their properties.

Learning Objectives:

1. Students will be able to list at least two physical properties of a substance.
2. Students will know that mixtures can be separated based on the physical properties of individual substances in the mixture.
3. Students will know that a conclusion is a claim supported by data.
4. Students will be able to distinguish between statements that are claim/data/neither.
5. Students will be able to match a claim with supporting data and realize when the data does not support the claim.
6. Students will be able to identify appropriate claims and data for a given data set.
7. Students will be able to list at least two ways that they behaved like scientists.

Classroom Teacher Responsibilities:

In order for SciTrek to be sustainable the program needs to work with teachers on developing their abilities to run student-centered inquiry-based science lessons in their own classroom. As teachers take over the role of SciTrek Lead, SciTrek will expand to additional classrooms. Even when teachers lead the modules in their own classroom, SciTrek will continue to provide volunteers and all of the materials needed to run the module. Below is a sample timeline for teachers to take over the role as the SciTrek lead.

1. Module 1 (year 1)
 - a. Classroom Teacher Co-Runs Group
 - i. Classroom teacher will co-run a group with a SciTrek volunteer. Groups contain ~ten students and will be subdivided into three small groups, ~four students, to perform experiments.
2. Module 2 (year 1)
 - a. Classroom Teacher Runs Group
 - i. Classroom teacher will run a group. Groups contain ~ten students and will be subdivided into three small groups, ~four students, to perform experiments.
3. Module 3 (year 2)
 - a. Classroom Teacher Runs Group and Starts Leading Class Discussions
 - i. Classroom teacher will run a group. Groups contain ~ten students and will be subdivided into three small groups, ~four students, to perform experiments.
 - ii. Classroom teacher will start leading parts of group discussions (e.g., conclusion activity, tie to standards, etc.)
4. Module 4 (year 2)
 - a. Classroom Teacher Leads the Class with Co-Lead Volunteer
 - i. Classroom teacher will be responsible for leading entire class discussions (e.g., conclusion activity, tie to standards, etc.).
 - ii. Classroom teacher will be responsible for time management.
 - iii. Classroom teacher will be responsible for overseeing volunteers and helping any groups that are struggling.
5. Any Additional Modules (year 3 and beyond)
 - a. Classroom Teacher Leads the Class
 - i. Classroom teacher will be responsible for leading whole class discussions (e.g., conclusion activity, tie to standards, etc.).
 - ii. Classroom teacher will be responsible for time management.
 - iii. Classroom teacher will be responsible for overseeing volunteers and helping any groups that are struggling.

SciTrek staff will be counting on teacher involvement after year one. Teachers should notify the SciTrek staff if they will not be present on any days of the module. Additional steps can be taken to become a SciTrek lead faster than the proposed schedule above. Contact scitrekadmin@chem.ucsb.edu to learn more.

In addition, teachers are required to come to UCSB for the volunteer orientation, typically one week prior to the start of the module. Contact scitrekadmin@chem.ucsb.edu for exact times and dates, or see our website at <http://www.chem.ucsb.edu/scitrek/module-times> under your class' module times. At the orientation teachers will meet the volunteers that will be helping in their classroom, go over module content, and learn their responsibilities during the module.

Prior to the Module (at least 1 week):

1. Come to the SciTrek orientation at UCSB.
2. Divide the class into three groups, ~ten students each. We find these groups work best when they are mixed levels and mixed language abilities. These larger groups of students will be further subdivided into three small groups, ~three students each, when performing experiments. You can also designate their small groups if desired. Send your class list showing the three large/small groups to scitreksadmin@chem.ucsb.edu one week before the start of the module so that we can have nametags/notebooks made for students.

During the Module:

If possible have a document camera available for the SciTrek lead every day of the module.

Day 1:

Have the students' desks/tables moved into three groups and cleared off. This ensures that each student has a desk during SciTrek activities.

Day 2-6:

Have the students' desks/tables moved into nine groups and cleared off. This ensures that each student has a desk during SciTrek activities and that students can begin with the module as soon as SciTrek arrives.

Day 7 and 8:

Have the students' desks/tables cleared off. The desks/tables do not need to be moved into groups.

Materials Used for this Module:

1. Crayons Crayola 8 count
2. Test tubes 25 x 150 mm (Fisher Part Number: 17-988-499)
3. Corks (Size 10) (Fisher Part Number: 07-781N)
4. Test tube stands (hand made by cutting a 2x4 into 15.5 cm long pieces and drilling four holes with a 1 in drill bit 2.5 cm deep along the center line of the block)
5. Nalgene graduated cylinders 10 ml (Fisher Part Number: 08-572-5A)
6. Chromatography paper (role 2 cm x 100 m (thickness 0.18mm) cut into 11.5 cm strips) (Fisher Part Number: S47087)
7. Other papers (all papers are cut into 2 cm x 10.5 cm strips)
Papers (coffee filter, construction paper, graph paper, newspaper, paper towel, and copy paper (white, pink, yellow, green, purple and blue))
8. Rulers (Office Max Part Number: 21215472)
9. MyChron Timers (Fisher Part Number: S65330)
10. Disposable pipets (droppers) (Fisher Part Number: 13-711-7M)
11. Markers
Mr. Sketch (red, orange, yellow light green, dark green, light blue, dark blue, purple, light pink, dark pink, black, and brown)
Crayola (red, yellow, green, blue, purple, black, and brown)
Expo-Overhead pens (red, yellow, green, blue, purple, black, and brown)
Sharpie (red, yellow, green, blue, purple, black, and brown)
Rose Art (red, yellow, green, blue, purple, black, and brown)
Other Black Pens (Bic, Dry Erase, and Paper Mate)

12. Water
13. Rubbing alcohol (RA)
14. White vinegar
15. Dish soap (without dilution the dish soap is too thick to be absorbed by the paper, therefore, a soap solution is made by mixing equal parts of water and dish soap)
16. 1 oz. Plastic cups (Smart and Final) labeled: water, RA, soap, and vinegar
17. Bags with 2 oz. of the following: baking soda, corn starch, salt, and sugar (both labeled with their names and A, B, C, and D) (Uline Part Number: S485)
18. Mixture bags (Mixture 1: bottle containing sand and water; Mixture 2: pieces of rough (2.5 cm x 2.5 cm) and coarse (1 cm x 2 cm) sand paper; Mixture 3: metal nuts (10-24) and nylon nuts (10-24) purchase from OSH; and Mixture 4: magnets (refrigerator magnets cut into 1 cm x 3 cm) and card stock (cut into 1 cm x 3 cm))

All printed materials used by SciTrek (nametags, material pages, poster parts, student notebooks, or instructions) can be made available for use and/or editing by emailing scitrekadmin@chem.ucsb.edu.

Day 1: Conclusion Assessment/Observations/Variables

Schedule:

Introduction (SciTrek Lead) – 2 minutes
 Conclusion Assessment (SciTrek Lead) – 10 minutes
 Observation Discussion (SciTrek Lead) – 2 minutes
 Observations (SciTrek Volunteers) – 24 minutes
 Variable Discussion (SciTrek Lead) – 5 minutes
 Variables (SciTrek Volunteers) – 12 minutes
 Wrap-Up (SciTrek Lead) – 5 minutes

Materials:

(3) Volunteer Boxes:

- | | | |
|---|--|---|
| <input type="checkbox"/> Student nametags | <input type="checkbox"/> Test tube with cork | <input type="checkbox"/> Black marker (Mr. Sketch) |
| <input type="checkbox"/> (12) Student notebooks | <input type="checkbox"/> Test tube stand | <input type="checkbox"/> Water (8 oz.) |
| <input type="checkbox"/> Volunteer instructions | <input type="checkbox"/> 10 ml Graduated cylinder | <input type="checkbox"/> (5) Boxes of crayons (8 colors only) |
| <input type="checkbox"/> Volunteer lab coat | <input type="checkbox"/> (2) Pieces of chromatography paper with pencil line drawn 2 cm above bottom | <input type="checkbox"/> Paper towels |
| <input type="checkbox"/> (2) Markers | <input type="checkbox"/> Dropper | <input type="checkbox"/> Timer |
| <input type="checkbox"/> Pencil | | <input type="checkbox"/> (2) Rulers |

Other Supplies:

- (3) Large group notepads (3) Trays Bucket

Lead Box:

- (5) Blank nametags (2) Pencils (2) Black markers (Mr. Sketch)
 (3) Extra student notebooks (2) Test tubes with corks Water (8 oz.)
 (35) Conclusion assessments (2) Test tube stands (5) Boxes of crayons (8 colors only)
 Time card (2) 10 ml Graduated cylinders
 Lead instructions (4) Pieces of chromatography paper with pencil line drawn 2 cm above bottom Paper towels
 Lead lab coat (2) Timers (2) Rulers
 (3) Markers (orange, green, blue) (2) Droppers

SciTrek Notebook Pages Used With Students:

OBSERVATIONS

Experimental set-up:
graduated cylinder with 2ml of water, paper with line on it at 2cm, test tube, cork, test tube holder, black Mr. Sketch pen, timer, 5 boxes of crayons

	Time 0	Time 1	Time 2
Time:	0 seconds	3min. 20sec.	7min. 15sec.
Picture:			
Observations/Measurements:	<ul style="list-style-type: none"> • small black dot on line (2cm) • strip put into test tube with water 	<ul style="list-style-type: none"> • dot turned into a smear (got larger) • black turned to blue and red 	<ul style="list-style-type: none"> • smear got bigger (3.0cm) • water went up paper • blue, pink, red, orange

2

Describe what happened during the experiment:

A black dot was put on a piece of paper with a Mr. Sketch pen. The paper was put into a test tube with 2ml of water. As time passed the black dot smeared upwards and turned from black into different colors (blue, orange, red). The paper also absorbed some of the water. The water went to 5.0 cm on the paper and the smear was 3.0 cm long.

Other Observations:

1. _____
2. _____
3. _____
4. _____

3

VARIBALES

What do you think might be affecting the smear?

Variable	How will the variable affect the smear height and color?
type of pen	I think that different black pens are made up of different colors so this might affect the colors seen in the smear.
type of paper	I think that if the paper is thicker then the dot will not travel as far up the paper.
type of liquid	I think that if the liquid is thicker it will also not be able to travel as high up the paper.
time	I think that the greater the amount of time the more separation there will be between the colors.
pen color	I think that different pen colors are also made of different colors and this might affect how high the colors travel.

4

Group Notebook Pages Used With Students:

<p style="text-align: right;">Teacher: _____ Volunteer: _____ color: _____</p> <p style="text-align: center;">Observations Experimental Set-up</p> <hr style="border: 0.5px solid black;"/> <hr style="border: 0.5px solid black;"/> <table border="1" style="width: 100%; border-collapse: collapse; margin-top: 10px;"> <thead> <tr> <th style="width: 15%;"></th> <th style="width: 20%;">Time 0</th> <th style="width: 20%;">Time 1</th> <th style="width: 20%;">Time 2</th> </tr> </thead> <tbody> <tr> <td style="text-align: center;">Time 1:</td> <td></td> <td></td> <td></td> </tr> <tr> <td style="text-align: center;">Observations/ Measurements</td> <td style="text-align: center;"></td> <td style="text-align: center;"></td> <td style="text-align: center;"></td> </tr> </tbody> </table>		Time 0	Time 1	Time 2	Time 1:				Observations/ Measurements				<p style="text-align: center;">Observations</p> <p>Describe what happened to the black dot:</p> <p style="text-align: right;">2</p>
	Time 0	Time 1	Time 2										
Time 1:													
Observations/ Measurements													

VARIABLES

What do you think might be affecting the smear?

Variable	How will the variable affect the smear height if color?

Set-Up:

SciTrek Lead:

If the classroom has a document camera, ask the teacher to use it to show the class question. If the classroom does not have a document camera, then write the class question on the board during the observation discussion.

SciTrek Volunteer:

Put your name, the teacher’s name, and your group color on the top of your group notepad.

As students are taking the conclusion assessment, walk around the room and quietly place the students’ nametags, which are in your group box, on each student’s desk.

Fill a 10 ml graduated cylinder with 2 ml of water. Place the test tube (with cork) in the test tube holder. Set the piece of chromatography paper (with the pencil line drawn at 2 cm from the bottom of the paper), test tube and holder, graduated cylinder with 2 ml of water, black Mr. Sketch marker, paper towel, timer, and ruler on the tray (see picture below). This set-up will be referred to as the experimental set-up.



Introduction (2 minutes – Full Class – SciTrek Lead):

“Hi, we are scientists from UCSB and we want to show you what we do as scientists. We will show you an experiment and then you can make observations, come up with a class question, and then design your own experiments to help answer the class question. We want to show you that you can do science and have fun.”

If you are a teacher that is leading the class tell your students that we are going to start a long-term science investigation and you have asked some scientists from UCSB to come and help. Allow the UCSB volunteers to introduce themselves and share their majors.

Conclusion Assessment (10 minutes – Full Class – SciTrek Lead):

As the students are taking the assessment, the volunteers should get the student nametags out of their group boxes and walk around the room locating their students. Have the volunteers quietly lay each student’s nametag on their desk. If students do not have their name on their paper remind them to do so. After nametags have been handed out volunteers should assemble the experimental set-up.

“Before we start with the module we will determine how your ideas on conclusions are developing.” Pass-out the conclusion assessment and tell students to fill out their name, teacher’s name, and date at the top of the assessment. Remind the students that it is important that they fill out this assessment on their own. For section one, read the instructions to the students. Then read each of the statements and tell the students to circle if the statement is a claim, data, or neither. As you are reading the statements walk around the room and verify that students have written their name on the top of the paper. For section two, read the instructions to the students and give them a few minutes to make matches between claims and data. You do not need to read the individual claim and data statements to the students. For section three, read the instructions to the students. Then read each of the statements. Tell students based on the results table to circle if the statement is an example of a claim, data, opinion, or if it is incorrect. You do not need to read or review the results tables with the students. Read the last question to the students and have them fill in the blank. When they are finished, collect the assessments and verify that the student’s name is on the top of the paper.

Observation Discussion (2 minutes – Full Class – SciTrek Lead):

Tell the students that scientists make lots of observations. Ask the class, “What is an observation? What are the types of things that you can record for an observation?” If they have trouble with this, show them an object and let them make some observations. Turn these observations into general features of an observation. Examples of possible general observations are: color, texture, size, weight, temperature, material, etc.

“In this experiment we are going to make observations of one strip of paper at three different times. The first time will be before we put the paper in the test tube with water and the other two will be after the paper is put in the test tube with water.” Remind them to make both written and illustrated observations in their SciTrek notebooks.

Tell the class they will now get in their groups and make observations. To determine their group they will need to look at the color of their nametag. Tell each colored group where to go and to bring a pencil with them.

If a student does not have a nametag write the student’s name on one of the extra nametags that are in the lead box using the color of marker of the group they will join.

Observations (24 minutes – Large Groups – SciTrek Volunteers):

Once the students come over to your group have them sit in boy/girl fashion. Make sure the table is set-up as described in the set-up section. Pass out a SciTrek notebook to each student. Before beginning the observations, make sure students fill out the entire front cover of their SciTrek notebook (except for group number and class question). Then have students turn to page 2 of their notebooks.

As a group, have the students come up with ~6 observations about the experimental set-up. Observations should be recorded on the large group notepad and then copied into student notebooks. Make sure that these observations include the following two observations 1) graduated cylinder with 2 ml of water and 2) piece of paper with a line 2 cm from the bottom. Give students ~ 2 minutes to generate a few unique observations to record in their own notebooks. Observations about the experimental set-up should not take longer than 5 minutes.

With the black Mr. Sketch marker, make a dot on the line of the chromatography paper (the smaller the dot the better the results). Have students look at the paper and tell them to remember what it looks like because this will be their time 0 observation. Pour the 2 ml of water into the test tube and place the chromatography paper into the test tube, cork the system, and start the timer. It is important that this is done prior to having students record their time 0 observations so that enough time passes (~ 3 minutes) between time 0 and time 1. Record 0 seconds under the time because this is when the strip of paper was put into the test tube. Then have students draw a picture of their observations of the paper at time 0. In addition, have students generate written observations/measurements and record these under the time 0 section of the chart on the group notepad. Then have the students copy this information into their SciTrek notebooks.

After students have completed their time 0 observations, record the time on the stopwatch, this will be time 1 (~3 minutes). DO NOT STOP THE TIMER OR TAKE THE STRIP OUT OF THE TEST TUBE AT TIME 1. Have the students draw a picture of what the paper looks like at time 1. Make sure students draw the

picture exactly like the strip of paper in front of them. A common mistake that students make is drawing the dye colors going down from the line instead of up from the line. Allow the students to use the provided crayons for their pictures. After their pictures are complete, make written observations as a group about time 1 and record these on the group notepad. Then have the students copy this information into their SciTrek notebooks.

After students have completed their time 1 observations, remove the paper and place it on a paper towel on the table. At the same time, stop the stopwatch. Record this time (~7 minutes) for time 2 on the group notepad. Ask the students, "What do they see now that will disappear by tomorrow?" They should respond with the water line. Tell the students in order to know the location of the water we use a pencil to trace the water level. Have the students draw a picture of the strip, including water line, for time 2. Then as a group, have the students come up with observations and record these observations on the group notepad. Make sure that in time 2 observations/measurements section of the chart that students record the height of the smear and the height of the liquid. Have the students copy these observations from the group notepad into their SciTrek notebooks. Filling in the chart should take your groups ~12 minutes.

Have students turn to page 3 of their notebooks. Have the students write a summary (multiple sentences) of what happened to the dot over the course of ~7 minutes. Probe them to write a better description if it takes them fewer than 7 minutes. Below are example questions to help students generate a more detailed description.

How did the dot/smear change over the course of time?

Did you see the same colors at each time?

Where was the water line in comparison to the height of the smear?

Why do you think the smear separated into different colors?

What do you think would happen if we left the paper in the water for a longer amount of time?

If students are struggling with writing a description have other group members tell you some of the main points that they included in their summary and write these on page 2 of the group notepad as bullet points. If the group can write the summary without the help of the group notepad, page 2 of the group notepad can be left blank.

If there is extra time have students come up with a few more general observations themselves and record these in their SciTrek notebook under other observations. Have students share these observations with the group. If you run out of time you can leave the other observation section blank because it is more important that they have an accurate description of what happened to the black dot.

An example of the group notepad/student notebook for all of the observations is seen below; feel free to deviate from the example.

Teacher: Ms. Johnson
Volunteer: Breanna
color: Orange

Observations

Experimental Set-up
graduated cylinder with 2ml of water, paper with line on it at 2 cm, black Mr. Sketch pen

	Time 0	Time 1	Time 2
Time:	0 seconds	3min. 20 sec.	7min. 15 sec.
Picture:			
Observations/Measurements	<ul style="list-style-type: none"> • small black dot on line (2cm) • strip put into test tube with water 	<ul style="list-style-type: none"> • dot turned into a smear (got longer) • black turned into blue and red 	<ul style="list-style-type: none"> • smear got bigger (3.0 cm) • water went up paper • blue, pink, red, orange

Observations

Describe what happened to the black dot:

- started as a black dot at 2cm
- put 2mL of water in test tube
- smeared 3.0 cm
- colors (blue, orange, pink, red)

OBSERVATIONS

Experimental set-up:
graduated cylinder with 2ml of water, paper with line on it at 2 cm, test tube, cork, test tube holder, black Mr. Sketch pen, timer, 5 boxes of crayons

	Time 0	Time 1	Time 2
Time:	0 seconds	3min. 20 sec.	7min. 15 sec.
Picture:			
Observations/Measurements	<ul style="list-style-type: none"> • small black dot on line (2cm) • strip put into test tube with water 	<ul style="list-style-type: none"> • dot turned into a smear (got longer) • black turned to blue and red 	<ul style="list-style-type: none"> • smear got bigger (3.0cm) • water went up paper • blue, pink, red, orange

Describe what happened during the experiment:

A black dot was put on a piece of paper with a Mr. Sketch pen. The paper was put into a test tube with 2mL of water. As time passed the black dot smeared upwards and turned from black into different colors (blue, orange, red). The paper also absorbed some of the water. The water went to 5.0 cm on the paper and the smear was 3.0 cm long.

Other Observations:

1. _____
2. _____
3. _____
4. _____

At the end of the observation section, each group will be asked to share one observation. Pick one student that will share an observation with the rest of the class.

Variable Discussion (5 minutes – Full Class – SciTrek Lead):

Have each group share one of their observations with the rest of the class.

Review with the class how the dot changed over time. Make sure that by the end of the discussion the students have identified that there was originally a black dot on the line. When the dot was placed in water the dot separated into several colors and the more time that passed the larger the separation became.

Ask the class what the most interesting observation was. They should reply that the black dot spread out into multiple colors. Tell the class we will then work together to answer the question, “What variables affect the black dot from spreading out into multiple colors?” More simply, “What variables affect the smear?” Write this question on the front page of the example notebook under the document camera and have students copy this question onto their notebook. Ask the class if they think that there is one variable that affects the smear or multiple variables. They should respond multiple variables. Tell the students this is why we will have to work as a class to answer the class question.

Tell students they are now going to discuss variables within their groups.

Ask the class what does the word “variable” mean to a scientist? What is the definition of a variable in science? Possible answer: variables are parts of the experiment that you can change.

Tell the class that they are going to think about variables in the experiment that they could change to help us answer the class question of: What variables affect the smear? In addition to generating variables they should think about why these variables might affect the outcome of the experiment. Ask the class to give you a variable that they think might affect the smear and have them tell you why or how they think that variable would affect the experiment. Then probe them on how they would design an experiment to test if this variable affected the smear. Finally, have the students make a prediction of the results of the experiment that they proposed.

Example: *Variable:* paper type
 Why this variable might affect the smear? Different papers might absorb different amount of liquid.
 How would you test this variable? Get different types of paper and put black dots on them and put them in water.
 Prediction: The more absorbent the paper the larger the smear would be.

Tell the students they will generate more variables and analyze them in their groups.

Variables (12 minutes – Large Groups – SciTrek Volunteers):

As a group, generate a variable and make a prediction about how it will affect the smear. Encourage and challenge students to explain why they think their prediction is correct and how this variable will affect the smear. Repeat this process two more times and record these ideas on the group notepad and then have students copy these into their notebooks. Then have students generate more variables and make

predictions about how different values of this variable will affect the smear and record their ideas in their notebook. Have students share these ideas with the group.

Prepare one student to share one of their variables and why they think it will affect the smear during the group discussion.

VARIABLES	
What do you think might be affecting the smear?	
Variable	How will the variable affect the smear height & color?
type of pen	I think that different black pens are made of different colors so this might affect the colors seen in the smear.
type of paper	I think that if the paper is thicker then the dot will not travel as far up the paper.
type of liquid	I think that if the liquid is thicker it will also not be able to travel as high up the paper.

VARIBALES	
What do you think might be affecting the smear?	
Variable	How will the variable affect the smear height and color?
type of pen	I think that different black pens are made up of different colors so this might affect the colors seen in the smear.
type of paper	I think that if the paper is thicker then the dot will not travel as far up the paper.
type of liquid	I think that if the liquid is thicker it will also not be able to travel as high up the paper.
time	I think that the greater the amount of time the more separation there will be between the colors.
pen color	I think that different pen colors are also made of different colors and this might affect how high the colors travel.

Wrap-Up (3 minutes – Full Class – SciTrek Lead):

Have one student from each group share a variable that they generated and why/how they think it will affect the smear. Make sure that students tell you their predictions about how different values of that variable will affect the smear. Challenge students to justify their thinking and explore with them how this might help them design an experiment. For example, if a students’ variable was paper type and they predicted that the more absorbent the paper the bigger the smear because the smear height never went higher than the liquid height ask them when they select their papers for their experiment if they would want all papers that absorbed the same amount of water. Then ask them to give you an example of a paper that absorbs a lot of water (paper towels) and a paper that does not absorb a lot of water (copy paper).

Tell the students that the next time we meet they will design an experiment to answer a question that they have about this experiment, which will help them learn about variables that might affect the smear.

Clean-Up:

Before you leave, have students attach their nametags to their notebooks and place them in the group box. Put the test tube with the liquid in it in the waste bucket. Do not leave the cork in the test tube or put it in the waste bucket; put the cork back in the group box. Bring all materials back to UCSB. In addition, put your lab coat back in your group box. If you would like to divide your large group (~10 students) into three smaller groups you can do this by writing a "1", "2", or "3" on the top each students notebook to designate their group. Make sure that the groups are made up of mixed gender and mixed ability students.

Day 2: Question/Experimental Set-Up/Procedure**Schedule:**

Introduction (SciTrek Lead) – 13 minutes
Question (SciTrek Volunteers) – 10 minutes
Experimental Set-Up (SciTrek Volunteers) – 15 minutes
Procedure (SciTrek Volunteers) – 14 minutes
Results Table (SciTrek Volunteers) – 5 minutes
Wrap-Up (SciTrek Lead) – 3 minutes

Materials:**(3) Volunteer Boxes:**

- | | | |
|---|---|---|
| <input type="checkbox"/> Student nametags | <input type="checkbox"/> Volunteer lab coat | <input type="checkbox"/> (3) Materials pages (group color/number indicated) |
| <input type="checkbox"/> Student notebooks | <input type="checkbox"/> Pencil | <input type="checkbox"/> Notepad |
| <input type="checkbox"/> Volunteer instructions | <input type="checkbox"/> (2) Red pens | |

Lead Box:

- | | | |
|--|---|--|
| <input type="checkbox"/> (5) Blank nametags | <input type="checkbox"/> Lead lab coat | <input type="checkbox"/> (2) Red pens |
| <input type="checkbox"/> (3) Extra student notebooks | <input type="checkbox"/> (3) Marker (orange, green, blue) | <input type="checkbox"/> (3) Materials pages |
| <input type="checkbox"/> Time card | <input type="checkbox"/> (2) Pencils | <input type="checkbox"/> Notepad |
| <input type="checkbox"/> Lead instructions | | <input type="checkbox"/> Example chromatography strip from Day 1 |

SciTrek Notebook Pages Used with Students:

Experimental Considerations:

- You will only have access to the materials on the materials page.
- You must make a measurement because you will make a bar graph of your data.
- The strips of paper cannot be in the liquid for more than 5 minutes.
- All strips of paper must be put into the liquid at the same time.
- You will only get four strips of paper for each experiment so you can only have four trials.

Changing Variable(s)

You will get to perform two experiments. For your first experiment decide which variable(s) (max three) that you would like to test. For each changing variable that you select state why you think that variable will affect the smear.

Changing Variable 1: color of pen
Discuss with your group how you think **changing variable 1** will affect the smear.

Changing Variable 2 (optional): type of liquid
Discuss with your group how you think **changing variable 2** will affect the smear.

Changing Variable 3 (optional): _____
Discuss with your group how you think **changing variable 3** will affect the smear.

QUESTION

Question our group will investigate:

- If we change the color of the pen and the type of liquid (insert each changing variable)
what will happen to the height and color of the smear (insert what you are measuring/observing)

SciTrek Member Approval: SJ

5

EXPERIMENTAL SET-UP

Determine the values of your changing variable(s) (ex: pen color) from the materials page and write the values (ex: trial A: blue) for your 4 trials under each strip of paper.

Changing Variable(s):

1) color of pen : red blue yellow green
 2) type of liquid : RA soap vinegar water
 3) _____

Controls (variables you will hold constant):
*Write at least 4 variables that you will hold constant and indicate the specific value you will use in your trials.

Test Tube / Original : _____ paper type / original
 temperature / RT time / 5 min
 liquid amount / 2 mL pen type / Mr. Sketch

SciTrek Member Approval: SJ

6

PROCEDURE

Procedure Note:
Make sure to include all values of your changing variable(s) in the procedure. (Example, for a group that decided to change pen color one step would be: put colored dot with a Mr. Sketch pen on original paper at 2 cm A) red, B) blue, C) green, and D) yellow.)

- Fill four test tubes with 2 ml of the following: A) RA B) soap C) vinegar D) water
- Put a colored dot with Mr. Sketch marker on each original paper at 2cm A) red B) blue C) yellow D) green
- Put papers in test tubes and then cork the test tubes
- Time for 5 minutes
- Take strips out of the liquid
- Measure and observe the height and color of the smear
- _____

SciTrek Member Approval: SJ

7

RESULTS Table

Fill out the chart for each of your trials. If one of the variables remains constant for all trials write the value in trial A and then draw a line through each box indicating that this variable is a control.

Variables:	Trial A	Trial B	Trial C	Trial D
Test Tube:	<u>Original</u>	_____	_____	_____
Time:	<u>5 minutes</u>	_____	_____	_____
Liquid Type:	<u>RA</u>	<u>soap</u>	<u>vinegar</u>	<u>water</u>
Liquid Amount:	<u>2 mL</u>	_____	_____	_____
Paper Type:	<u>original</u>	_____	_____	_____
Pen Color:	<u>red</u>	<u>blue</u>	<u>yellow</u>	<u>green</u>
Pen Type:	<u>Mr. sketch</u>	_____	_____	_____
Initial Dot Height:	<u>2 cm</u>	_____	_____	_____
Predictions:	Trial A	Trial B	Trial C	Trial D
Put a "T" in the trial that will give the tallest smear and an "S" in the trial that will give the shortest smear.		<u>S</u>		<u>T</u>
Data:	Trial A	Trial B	Trial C	Trial D
Final Observations/Measurements:	Liquid height			
	Smear height			
	Color			

SciTrek Member Approval: _____

8

Set-Up:

SciTrek Lead:

Make sure that each of the larger groups (~ten students) are divided into three smaller groups (three/four students). When dividing the students into smaller groups, try to form mixed gendered/ability groups. Before arriving to class indicate each student's group by writing their group number on the front cover of their SciTrek notebook.

If the classroom has a document camera, ask the teacher to use it for the question, experimental set-up, and results examples (pages 5, 6, and 8). If the classroom does not have a document camera, then tape the example poster-size notebook pages to the front board.

SciTrek Volunteer:

While the SciTrek lead is giving the introduction pass out the SciTrek notebooks/nametags to the students in your group. After the introduction, when students come to your group make sure that students with the same number are sitting next to each other. Have materials pages with group numbers and colors filled out to give students after they complete their questions. Have a red pen available to approve students' question, experimental set-up, and procedure (pages 5, 6 and 7).

Introduction (13 minutes – Full Class – SciTrek Lead):

While you are doing the introduction have the SciTrek volunteers pass out the SciTrek notebooks/nametags to the students in their group.

Ask the students what they did during the last meeting with SciTrek, and show them the example strip of chromatography paper to help remind them. They should reply that they put a black dot on a piece of paper and observed that the dot separated into different colors when the paper absorbed water. Ask the class if they remember the class question they decided to investigate. They should reply, "What variables affect the smear?" Tell students that one way scientists answer questions is by performing experiments; today they will design an experiment to help answer the class question. Ask the class if they think there are multiple variables that could affect the smear. They should respond that there probably are multiple variables. Therefore, each group will generate a smaller question to investigate. Once we put all the groups' research together we should be able to answer the class question. After each group has generated a question they will determine the values of their changing variable(s) and then determine their controls; this is sometimes known as an experimental set-up. After they have generated their experimental set-up they will write a list of steps to follow during their experiment, also known as a procedure. Tell the students that they need to keep a few things in mind while designing their experiment.

Experimental Considerations:

1. You will only have access to the materials on the materials page.
2. You must make a measurement because you will make a bar graph of your data.
3. The strips of paper cannot be in the liquid for more than 5 minutes.
4. All strips of paper must be put into the liquid at the same time.
5. You will only get four strips of paper for each experiment so you will only have four trials.

Tell the students we are now going to generate an example question/experimental set-up together and that you will write it in an example notebook so that they will be able to refer back to it when they are completing the process themselves. Make sure that students DO NOT fill out the example question/experimental set-up in their notebooks.

Ask the students for one variable that they think will affect the smear (example answer: pen color). For whatever variable is chosen have the students tell you how they think this variable will affect the smear (example answer: the black pen separated into different colors therefore, maybe different pens color separate into different colors. In addition, I think that the height of the smear will be approximately the same height as the original smear because I cannot see how pen color would change the height.) Push students to make predictions on both color and height of the smear. Then ask them if we want a narrow or a wide range of values for this variable and why (example answer: we would want as many different pen colors as possible so that we could see if they all were different colors). Then write down the suggested variable on the example notebook (page 5) under the document camera. Ask the students if they think any other variables will affect the smear (example answer: liquid type). Then ask them to tell you why they think this variable will affect the smear (example answer: the thicker the liquid the less it would absorb up the paper and the smaller the smear would be. However, I do not think that the liquid type would affect the colors that come out of the dot because the colors are a result of the ink and not the liquid). Finally, ask them if we want a narrow or a wide range of values for this variable and why (example answer: we would like a wide range of liquids to see the effects easier). Then write down the second changing variable on the example notebook. Do not fill in the blank for the third changing variable or the blanks explaining why they think each changing variable would affect the smear. Tell the students when they are going through this process in their small groups they can design an experiment that has one, two, or three changing variable(s).

Show students how to insert their changing variable(s) into the question frame to find the question they will investigate. For the example discussed above the question would be: If we change the pen color and liquid type what will happen to the height and color of the smear? Explain to students that many times when there is a large question, like our class question, scientists break it down into smaller questions that individual scientists can investigate and then they compile their work to answer the large question.

Tell them once they have determined their question and have approval, their SciTrek volunteer will give them a materials page for determining the values of their changing variable(s) and controls. Ask students if they know how scientists define controls. Make sure that by the end of the conversation students understand that controls are variables that are held constant during an experiment. For example, if they used Mr. Sketch pens for all the strips then one of their controls would be pen type. These controls can be different than the original experiment.

Show the students the materials page and have them help you decide the values for the changing variables. If they choose a value contrary to their experimental design, question them on their reasoning. For example, if they said they wanted to use a wide range of pen colors and they picked red, light pink, dark pink, and orange ask them if this values would allow them to best answer their question. Then allow them to change their values if needed. See the example notebook below for possible class choices.

Write these values on the example notebook. In addition, show them how to select the materials that they will need on the materials page.

Once they have determined the values of their changing variables have them generate a list of controls. First generate five things that could be controls and then show them the materials page again and have them fill in the values. For example the controls could be: pen type, time, dot height, liquid amount, paper type. Write these variables on the left side of each of the slashes in the example notebook. Then have them use the materials page to determine the exact values that they will use for each control. After each value is selected ask the class if having that value would make it easier or harder for them to find the answer to their question. For instance, if one of their constants was time / 1 minute, ask them will it be harder or easier to determine if pen color and liquid type is affecting the color and the height of the smear. Hopefully, they will realize that having a time of 1 minute would make it harder to analyze their results because their smear would not have time to travel up the paper. If they realize this might make their experiment more difficult, allow them to change their selection to 5 minutes. Record all of the values of the controls on the right side of the slash. Also, fill out the materials page appropriately. See example notebook/materials page below for possible class choices.

Tell students that once they have their experimental set-up complete they will have it approved by their SciTrek volunteer. Next, they will write a procedure that they will be able to follow next time. When writing their procedure they should write all the values of their changing variable(s). Show students the example procedure step on page 7 of their notebook (put dot on each strip of paper A) red, B) blue, C) green, and D) yellow). Once their procedure is completed they will get it approved by a SciTrek volunteer and then they will fill out their results table so they will be ready to start their experiment during SciTrek's next visit.

Put the results table (page 8) under the document camera. Tell students when they have a constant variable, or a control, they will just write it in trial A and then draw a line through the remaining trials. Put one or two of the class controls in the table. However, if a variable is changing then the values need to be written in all of the boxes. Record the class changing variable values in the appropriate boxes. Tell students the last thing they will need to do is make predictions about the smear height at the end of the experiment. Tell them they can label the trial that they think will result in the tallest smear with a "T" and the trial they think will result in the shortest smear with a "S," leaving the other trials blank. If they think that all the smears will be the same height them can write "same" across all the boxes. Show them where to fill this in on the results table but do not fill the predictions in for the class.

Tell students that they will now divide into groups and design their experiments. Leave the question/experimental set-up on the document camera so that students can refer back to it as they design their experiments. Below is an example of what should be filled in the example notebook during the introduction. Note that several sections are left blank.

Experimental Considerations:

1. You will only have access to the materials on the materials page.
2. You must make a measurement because you will make a bar graph of your data.
3. The strips of paper cannot be in the liquid for more than 5 minutes.
4. All strips of paper must be put into the liquid at the same time.
5. You will only get four strips of paper for each experiment so you can only have four trials.

Changing Variable(s)

You will get to perform two experiments. For your first experiment decide which variable(s) (max three) that you would like to test. For each changing variable that you select state why you think that variable will affect the smear.

Changing Variable 1: color of pen

Discuss with your group how you think **changing variable 1** will affect the smear.

Changing Variable 2 (optional): type of liquid

Discuss with your group how you think **changing variable 2** will affect the smear.

Changing Variable 3 (optional): _____

Discuss with your group how you think **changing variable 3** will affect the smear.

QUESTION

Question our group will investigate:

- If we change the color of the pen and the type of liquid
insert each changing variable

what will happen to the height and color of the smear?
insert what you are measuring/observing

SciTrek Member Approval _____

Color: Drifting
Group Number: 3

MATERIALS

You will only have access to the following materials. Circle the materials that you need for your experiment.

General Materials:

- test tubes with cork
- test tube rack
- paper
- rulers
- 2 droppers
- 2 graduated cylinders

Types of Liquid:

- rubbing alcohol (Kk)
- soap
- vinegar
- water

Types of Paper: You may only have 6 strips of paper (each paper that you circle get the amount of strips that you need on the line. All of the numbers should add up to 6.)

- original paper (this is the paper that was used on the first day in the group experiment)
- white filter
- construction paper
- copy paper
- graph paper
- newspaper
- paper towel
- copy paper pink
- copy paper yellow
- copy paper green
- copy paper purple
- copy paper blue

Types of Pens/Colors:

- Mr. Sketch (original)
- Expo
- Sharpie
- Rose Art
- blue
- dark green
- light pink
- orange
- light blue
- dark pink
- purple
- black (original)
- yellow
- dark blue
- light green
- purple
- brown
- green
- brown
- green
- brown
- green
- brown
- green
- brown
- green
- brown
- green
- brown

Types of Black Pen:

- Mr. Sketch
- Expo
- Rose Art
- Crayola
- Mr. Sketch
- Sharpie
- Dry Erase
- Paper Mate

EXPERIMENTAL SET-UP

Determine the values of your changing variable(s) (ex: pen color) from the materials page and write the values (ex: trial A: blue) for your 4 trials under each strip of paper.



Changing Variable(s):

- 1) color of pen: red blue yellow green
- 2) type of liquid: RA soap vinegar water
- 3) _____

Controls (variables you will hold constant):

*Write at least 4 variables that you will hold constant and indicate the specific value you will use in your trials.

- Test Tube / Original
- temperature / RT
- liquid amount / 2 mL
- paper type / original
- time / 5 min
- pen type / Mr. Sketch

SciTrek Member Approval _____

RESULTS Table

Fill out the chart for each of your trials. If one of the variables remains constant for all trials write the value in trial A and then draw a line through each box indicating that this variable is a control.

Variables:	Trial A	Trial B	Trial C	Trial D
Test Tube:	Original	—————		
Time:	5 minutes	—————		
Liquid Type:	RA	soap	vinegar	water
Liquid Amount:	2 mL	—————		
Paper Type:	original	—————		
Pen Color:	red	blue	yellow	green
Pen Type:	Mr. Sketch	—————		
Initial Dot Height:	2cm	—————		
Predictions:	Trial A	Trial B	Trial C	Trial D
Put a "T" in the trial that will give the tallest smear and an "S" in the trial that will give the shortest smear.				
Data:	Trial A	Trial B	Trial C	Trial D
Final Observations/Measurements:	Liquid Height:			
	Smear Height:			
	Time:			

Question (10 minutes – Small Groups – SciTrek Volunteers):

Have students get into their small groups. Have them determine the variable(s) that will be their changing variable(s). If they have only one changing variable do not encourage them to have more and if they have two/three changing variables do not encourage them to have fewer. Students will analyze their data and then perform an additional experiment to correct any mistakes that they made on their first experiment. Then have each group discuss briefly why/how they think each changing variable will affect the smear. Once students have chosen their changing variable they can fill out their question. When you sign off on their question give them a materials page with their group color and number designated in the upper right hand corner. An example notebook is seen below.

Pick one student to share the question that their group is going to investigate during the class discussion.

Experimental Considerations:

1. You will only have access to the materials on the materials page.
2. You must make a measurement because you will make a bar graph of your data.
3. The strips of paper cannot be in the liquid for more than 5 minutes.
4. All strips of paper must be put into the liquid at the same time.
5. You will only get four strips of paper for each experiment so you can only have four trials.

Changing Variable(s)

You will get to perform two experiments. For your first experiment decide which variable(s) (max three) that you would like to test. For each changing variable that you select state why you think that variable will affect the smear.

Changing Variable 1: color of pen
Discuss with your group how you think **changing variable 1** will affect the smear.

Changing Variable 2 (optional): type of liquid
Discuss with your group how you think **changing variable 2** will affect the smear.

Changing Variable 3 (optional): _____
Discuss with your group how you think **changing variable 3** will affect the smear.

QUESTION

Question our group will investigate:

- If we change the color of the pen and the type of liquid
Insert each changing variable

what will happen to the height and color of the smear
Insert what you are measuring/observing

SciTrek Member Approval: ST

5

Experimental Set-Up (15 minutes – Small Groups – SciTrek Volunteers):

Have groups determine the values for their changing variable(s) and record them in their experimental set-up on page 6. Encourage students to determine their controls before determining the values of their controls. Ask students to justify the values that they have chosen for their controls and if these values will make it easier or harder to answer their question. In addition, collect the materials page and make sure that it is filled out correctly and completely. This is essential for starting their experiments during the next SciTrek visit. An example of the experimental set-up and materials pages are seen below.

EXPERIMENTAL SET-UP

Determine the values of your changing variable(s) (ex: pen color) from the materials page and write the values (ex: trial A: blue) for your 4 trials under each strip of paper.

A

B

C

D

Changing Variable(s):

1) color of pen : red blue yellow green

2) type of liquid : RA soap vinegar water

3) _____

Controls (variables you will hold constant):
*Write at least 4 variables that you will hold constant, and indicate the specific value you will use in your trials.

Test Tube / Original	paper type / original	temperature / RT	time / 5 min.
liquid amount / 2 mL	pen type / Mr. Sketch		

SciTrek Member Approval

MATERIALS

Color: Drainage
Group Number: 3

You will only have access to the following materials. Circle the materials that you need for your experiment.

General Materials:
 test tubes with cork
 test tube rack
 paper
 rulers
 2 droppers
 2 graduated cylinders

Types of Liquid:
 rubbing alcohol (RA)
 soap
 vinegar
 water

Types of Paper: You may only have 4 strips of paper (each paper that you circle get the number of strips that you need on the line. All of the numbers should add up to 4.)
 original paper (this is the paper that was used on the first day in the group experiment) 1
 white filter _____
 construction paper _____
 graph paper _____
 copy paper pink _____
 copy paper purple _____
 newspaper _____
 copy paper yellow _____
 copy paper blue _____
 paper towel _____
 copy paper green _____

Types of Pen/Colors:
 Mr. Sketch (original)
 Expo
 Rose Art
 Sharpie
 Expo
 Rose Art
 Sharpie

orange	light blue	yellow	light green
dark green	dark pink	black (original)	purple
light pink	dark purple	black	green
orange	purple	yellow	green
blue	purple	black	green
orange	purple	yellow	green
blue	purple	black	green
orange	purple	yellow	green
blue	purple	black	green

Types of Black Pen:
 Mr. Sketch
 Expo
 Rose Art
 Crayola
 Mr. Sketch
 Sharpie
 Dry Erase
 Paper Mate

Procedure (14 minutes – Small Groups – SciTrek Volunteers):

After each small group has filled out their experimental set-up they can start on their procedure (page 7). Keep procedures as brief as possible while still conveying the pertinent information about the experiment. Make sure that you have students include all changing variable values in the procedure. For example, if pen color was one of the changing variables one step of the procedure would be, “put colored dot with Mr. Sketch pen on original paper at 2 cm A) red, B) blue, C) green, and D) yellow.” Some groups may struggle with writing a procedure. You can have these groups dictate each step while you transcribe them onto a notepad found in the group box. Give this sheet to the students to copy into their notebooks. Once the students have their procedure written in their notebooks they should raise their hands to get approval by their SciTrek volunteer. An example procedure can be found below.

PROCEDURE

Procedure Note:
 Make sure to include all values of your changing variable(s) in the procedure. (Example, for a group that decided to change pen color one step would be: put colored dot with a Mr. Sketch pen on original paper at 2 cm A) red, B) blue, C) green, and D) yellow.)

1. Fill four test tubes with 2ml of the following:
 A) RA B) soap C) vinegar D) water
2. Put a colored dot with Mr. Sketch marker on each original paper at 2cm A) red B) blue C) yellow D) green
3. Put papers in test tubes and then cork the test tubes
4. Time for 5 minutes
5. Take strips out of the liquid
6. Measure and observe the height and color of the smear
7. _____

SciTrek Member Approval

7

RESULTS Table

Fill out the chart for each of your trials. If one of the variables remains constant for all trials write the value in trial A and then draw a line through each box indicating that this variable is a control.

Variables:	Trial A	Trial B	Trial C	Trial D
Test Tube:	original	—————		
Time:	5 minutes	—————		
Liquid Type:	RA	soap	vinegar	water
Liquid Amount:	2 mL	—————		
Paper Type:	original	—————		
Pen Color:	red	blue	yellow	green
Pen Type:	Mr. sketch	—————		
Initial Dot Height:	2cm	—————		
Predictions:	Trial A	Trial B	Trial C	Trial D
Put a "T" in the trial that will give the tallest smear and an "S" in the trial that will give the shortest smear.		S		T
Data:	Trial A	Trial B	Trial C	Trial D
Final Observations/Measurements:	Liquid height			
	Smear height			
	Other			

8

Results Table (5 minutes – Small Groups – SciTrek Volunteers):

Have groups fill in their results table. Make sure that for their controls they only write the value of the control in trial A and then draw a line through the remaining trials. For the changing variable(s) they need to write the value of the variable in each of the boxes. When students have finished, have them make predictions about the final height of the smear. Have them write a "T" in the blank of the smear that they think that will be the tallest and an "S" in the blank of the smear that they think will be the shortest. This will leave two of the blanks empty. If they think that all trials will be the same height have the write "same" over all the boxes. Try to question each group on their thought process behind their predicted order. See example notebook above.

If there is extra time have the group close their notebooks and explain to each other what they will do for their experiment.

Wrap-Up (3 minutes – Full Class – SciTrek Lead):

Have one group from each of the large groups share what question they are investigating. Tell the students that on the next SciTrek visit they will start their experiments.

Clean-Up:

Before you leave, have students attach their nametags to their notebooks and place them in the group box. Place the materials pages on top of the notebooks in your group box. Bring all materials back to UCSB. In addition, put your lab coat back in your group box.

Day 3: Experiment/Conclusion Activity
Schedule:

Introduction (SciTrek Lead) – 8 minutes
 Experiment (SciTrek Volunteers) – 20 minutes
 Bar Graph (SciTrek Volunteers) – 10 minutes
 Conclusion Activity (SciTrek Lead) – 20 minutes
 Wrap-Up (SciTrek Lead) – 2 minutes

Materials:

(3) Volunteer Boxes:

- | | | |
|---|--|--|
| <input type="checkbox"/> Student nametags | <input type="checkbox"/> (2) Red pens | <input type="checkbox"/> Vinegar |
| <input type="checkbox"/> Student notebooks | <input type="checkbox"/> Notepad | <input type="checkbox"/> Water |
| <input type="checkbox"/> Volunteer instructions | <input type="checkbox"/> (3) Test tube stands | <input type="checkbox"/> Soap |
| <input type="checkbox"/> Volunteer lab coat | <input type="checkbox"/> (12) Small cups (labeled with liquid types) | <input type="checkbox"/> Rubbing alcohol |
| <input type="checkbox"/> (3) Pencils | | <input type="checkbox"/> Paper towels |
| | | <input type="checkbox"/> (6) Rulers |

(3) Ziploc Bags labeled group 1, 2, and 3 each with the following:

- | | | |
|--|--|--------------------------------------|
| <input type="checkbox"/> (2) 10 ml Graduated cylinders | <input type="checkbox"/> (2) Droppers | <input type="checkbox"/> Paper towel |
| <input type="checkbox"/> (4) Corks | <input type="checkbox"/> Requested pens | <input type="checkbox"/> Timer |
| | <input type="checkbox"/> Requested strips of paper | |

Other Supplies:

- | | |
|--|--|
| <input type="checkbox"/> Box of test tubes | <input type="checkbox"/> Bucket with lid |
|--|--|

Lead Box:

- | | | |
|---|---|---|
| <input type="checkbox"/> (3) Student notebooks | <input type="checkbox"/> (12) Small cups (labeled with liquid types) | <input type="checkbox"/> (5) Bags of different colored markers (Mr. Sketch, Crayola, Expo, Sharpie, and Rose Art) |
| <input type="checkbox"/> Lead instructions | <input type="checkbox"/> (4) 10 ml Graduated cylinders | <input type="checkbox"/> Vinegar |
| <input type="checkbox"/> Time card | <input type="checkbox"/> Bag of chromatography paper (minimum 30) | <input type="checkbox"/> Water |
| <input type="checkbox"/> Filled out results table | <input type="checkbox"/> Bag with the 11 different kinds of papers (minimum 20 each) | <input type="checkbox"/> Soap |
| <input type="checkbox"/> Lead lab coat | <input type="checkbox"/> Bag of 8 different black pens (Bic, Crayola, Dry Erase, Expo, Mr. Sketch, Paper Mate, Rose Art, and Sharpie) | <input type="checkbox"/> Rubbing alcohol |
| <input type="checkbox"/> (2) Pencils | | <input type="checkbox"/> (6) Rulers |
| <input type="checkbox"/> (2) Red pens | | <input type="checkbox"/> (2) Timers |
| <input type="checkbox"/> Notepad | | <input type="checkbox"/> Paper towels |
| <input type="checkbox"/> (2) Test tube stands | | |
| <input type="checkbox"/> (10) Corks | | |
| <input type="checkbox"/> (8) Droppers | | |

SciTrek Notebook Pages Used With Students:

RESULTS Table

Fill out the chart for each of your trials. If one of the variables remains constant for all trials write the value in trial A and then draw a line through each box indicating that this variable is a control.

Variables:	Trial A	Trial B	Trial C	Trial D	
Test Tube:	Original	—————→			
Time:	5 minutes	—————→			
Liquid Type:	RA	soap	vinegar	water	
Liquid Amount:	2 mL	—————→			
Paper Type:	original	—————→			
Pen Color:	red	blue	yellow	green	
Pen Type:	Mr. Sketch	—————→			
Initial Dot Height:	2 cm	—————→			
Predictions:					
Put a "T" in the trial that will give the tallest smear and an "S" in the trial that will give the shortest smear.		S		T	
Data:					
Final Observations/Measurements:	Liquid Height:	4 cm	6.5 cm	9.5 cm	5 cm
	Smear Height:	2 cm	4.5 cm	7.5 cm	3 cm
	Other:	red	blue, green, yellow	yellow	green, yellow, blue

11

RESULTS Graph

Record what you measured for each trial (example: smear height) on the y-axis (vertical axis) and each of your changing variable(s) values (example: red/water) on the x-axis. There may be blank lines on your x-axis. Determine an appropriate scale and record your data from each of your trials on the chart.

Changing Variable(s):

color of pen: red blue yellow green
 liquid type: RA soap water vinegar

Comparing Predictions and Actual Data
 Look at your predicted and actual smear heights (page 8). Were you surprised by your results why or why not? Discuss your actual and predicted data with your group as well as how you will use what you learned to design your next experiment.

12

SCIENTIFIC PRACTICES
Conclusions

1. Directions: Fill in the missing definition.

- Conclusion:** A claim supported by data.
 - Claim:** A statement that can be tested. The explanation of the data, the first part of a conclusion.
 - Example: Cats on average weigh less than dogs
 - Data:** Evidence collected from experiment(s) (measurements or observations), the second part of a conclusion.
 - Example: the average weight of a dog is 14 kg and the average weight of a cat is 5 kg.

2. Directions: Circle CLAIM if the statement is a claim. Circle DATA if the statement is data. Circle NEITHER if the statement is neither claim nor data.

- McDonalds served 100 customers and Taco Bell served 75. Claim Data Neither
- blue is the best color. Claim Data Neither
- butterflies that are larger than 15 cm are attracted to bright colors. Claim Data Neither
- ice has been observed to float on water. Claim Data Neither
- more people buy pizza than hamburgers. Claim Data Neither
- the average male blue whale weighs 100 tons, while the average female blue whale weighs 135 tons. Claim Data Neither

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SCIENTIFIC PRACTICES
Conclusions

3. Directions: Draw a line connecting claims with the correct data. If there is no data that supports the claim, do not draw a line.

Claim	because	Data
1. People read more from electronic devices than books		a. Sooy TVs give off 20 lumens of light and Samsung TVs give off 10 lumens of light.
2. Sooy TVs are brighter than Samsung TVs		b. when blue and red paint were mixed together the paint was observed to turn purple.
3. The color purple is made from blue and red		c. wind turbines produce 6,000 MW of energy and solar panels produce 5,000 MW of energy in California.
4. Wind turbines produce less energy than solar panels in California		d. the speed of light is measured to be 3×10^8 m/s.

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Set-Up:

SciTrek Lead:

If the classroom has a document camera, ask the teacher to use it for the graph example (page 9) and the conclusion activity (pages 10 and 11). If the classroom does not have a document camera, then tape the example poster-size notebook pages to the front board. Make sure that the results table (the example that will be used to show students how to make their graphs) is completed prior to entering the classroom.

SciTrek Volunteer:

While the SciTrek lead is giving the introduction pass out the SciTrek notebooks/nametags to the students in your group. Once the notebooks/nametags are passed out place the test tubes in the test tube rack and pour all of the liquids that your groups need into the small cups. Have all supplies ready so that you can set them out as soon as your groups are ready to start. After the introduction, when students come to your group make sure that students with the same number are sitting next to each other.

Introduction (8 minutes – Full Class – SciTrek Lead):

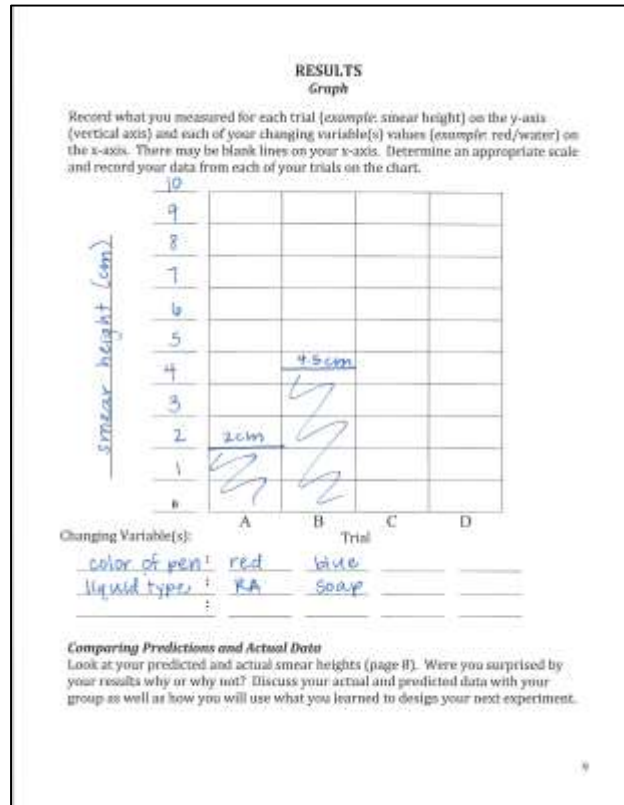
While you are doing the introduction have the SciTrek volunteers pass out the SciTrek notebooks/nametags to the students in their group and then set-up for the experiment

Ask the class, “What is the class question that we are investigating?” The students should reply, “What variables affect the smear?” Tell them that today they will start their experiment to answer this question. However, before they can start their experiment they need to have their results table completed (most students will have completed this the day before). Once this is finished they can raise their hands and they will receive their experimental supplies from their SciTrek volunteer.

Tell students that when they record their data they will make two measurements: the height of the liquid and the height of the smear. In addition, they will record any other observations such as the colors that are observed in each smear.

Tell the students that once they have collected their data they will plot their measurements on a graph (page 9). Tell the students you will show them how to make a graph using your data but they should not copy this data into their notebooks; they will graph their own data. Tell the students that your question was, “If we change the pen color and the liquid type what will happen to the height of the smear?” Therefore, you will plot smear height (when students do their actual experiment if none of the dots moved they can go back and modify their question to be what will happen to the height of the liquid and plot liquid heights instead of smear heights). Ask the students what they think we should count by on the y-axis (ones). Put the numbers on the graph, making sure that they know to start counting at zero. Make sure that you completely fill out the y-axis numbers to the top of the graph and do not stop numbering after you have passed the largest number that you measured. Ask students what we are plotting on the y-axis (the smear height). Fill this in on the y-axis title line. On the x-axis they need to record the values of the changing variable(s) for the experiment. Mention that by plotting their changing variable(s) on the x-axis they can see how these variables affected the height of the smear. If they had more than one changing variable they should record all of the changing variables. Fill in the two changing variables for the first two trials. Tell the students that we now need to transcribe our data on the graph. Put your finger at zero and ask the students to stop you once you reach the appropriate level for each of the changing variables. Once you have reached the level, draw the line, write the number value over the line,

and quickly shade below the line. Tell students to look how fast that you filled in the chart and challenge them to fill in their graph faster than you, when the plot their data. Repeat this process for one other data point. Thus you will graph the first two data points, leaving data points three and four blank. Leave the graph and the data table on the document camera for the class to refer back to while they are doing their experiments. An example graph is seen below.



Tell the students that they will now start their experiments and as soon as they are done they can graph their results.

Experiment (20 minutes – Small Groups – SciTrek Volunteers):

If you still have groups that have not finished their results table, help them complete it before giving them their requested materials. All student materials should be in your group box and all extra materials will be in the lead box. Make sure that students label their strips A, B, C, and D (in pencil at the top of each strip) so they can tell them apart later. As soon as students are done with their liquids, remove the liquids, graduated cylinders, and droppers and put them in the bucket (please do not put trash in the bucket). It is important to do this as soon as possible so students do not play with or spill their liquids. When the experiment is finished, place all test tubes in the bucket and put the corks, test tube rack, timers, and pens in your group box. If your group has things under control please help other groups. Once students have finished their experiments, they can record their findings. Make sure that groups trace the liquid line (with pencil) onto their strips so they can easily see/measure it later if needed. Once students have finished their measurements, place the chromatography strips in the Ziploc bag with their group number on it (this is the same bag that their supplies came in). Once a group has finished they can move on to graphing their results.

Bar Graph (10 minutes – Small Groups – SciTrek Volunteers):

Help students fill out their bar graphs (page 9). If students pick systems in which the dot did not smear they can go back to page 5 and revise their question to, “What will happen to the height of the liquid?” from, “What will happen to the height of the smear?” Then they will be able to plot liquid height instead of smear heights. Once they have graphed their values, make sure that they write the number on top of their line so that it is easy to discern the value.

Once students have completed their bar graph, have them use their results table to compare their predicted and actual smear heights and discuss why they think their results were different and how this experiment will help them with their next experiment.

If students finish early they can look at the conclusion activity on page 10 of their notebooks and start labeling each statement as claim, data, or neither.

Conclusion Activity (20 minutes – Full Class – SciTrek Lead):

Tell the students to turn to page 10 in their notebooks. Put a blank notebook under the document camera and turn to page 10. Mention that before they analyze their graphs and draw conclusions, it is important that they recognize and understand other’s conclusions.

Ask the class, “What is a conclusion?” After listening to the student’s answers make sure that the students understand that a conclusion is **a claim supported by data**. Write this definition on page 10 of the example notebook for the students to copy.

Tell the students that in order to make a conclusion we need to make sure that we understand the difference between claim and data. First, read the claim definition and the example. Tell the students that a claim is something that we can verify by testing. Have the class generate approximately four examples of statements that are claims. After a student suggests a possible claim ask the class if the possible claim can be verified by testing. Have the class hold their thumb up if it is a claim and down if it is not. Then ask someone else in the class to propose how you would test this claim. Several examples are seen below.

Examples: Claim: rabbits are faster than mice
Test: time rabbits and mice running a certain distance

Claim: giraffes are taller than horses
Test: measure the height of a horse and a giraffe

Claim: the average watermelon weighs more than the average pumpkin
Test: weigh pumpkins and watermelons to determine the average weight

Next, read the definition of data and the example aloud. Note that the example data supports the example claim, therefore, by combining the two statements, a conclusion can be formed. This conclusion would be: Cats, on average, weigh less than dogs because the average weight of a dog is 14 kg and the average weight of a cat is 5 kg. Tell students that data often contains a numerical measurement such as a height (5 m) or a weight (20 kg). Ask the students if data has to contain a numerical measurement. Explain that data can also be in the form of observations. For example, plants are observed to have

greener leaves when in direct light than indirect light. When you want to identify if a statement is data look for measurements or words such as *recorded* or *observed* that allow you to know that an experiment was performed.

Ask the students if all statements have to be either a claim or data. Lead students into realizing that some statements are neither a claim nor data; a common example of a statement that is not a claim or data is an opinion statement. Have students generate approximately four examples of opinion statements that are neither claims nor data.

Example: Watermelons taste better than pumpkins.
Rabbits are cuter than cats.

Read the directions to part 2 aloud to the class. Then work on the activity as a class. Tell the students to look for clues in the sentences to identify the statement as claim, data, or neither. Have the students share what they think is the correct answer and why. Have the class vote using thumbs up/thumbs down if they agree/disagree with the student's reasoning. After the class has come to agreement, mark the correct answer on the example notebook for students to copy. If the statement is a claim, have the students speculate what data they would need to collect in order to make a conclusion. If the statement is data, have the students generate a claim that could be supported by that data.

SCIENTIFIC PRACTICES
Conclusions

1. Directions: Fill in the missing definition.

- **Conclusion:** A claim supported by data.
 - **Claim:** A statement that can be tested. The explanation of the data, the first part of a conclusion.
 - Example: Cats on average weigh less than dogs
 - **Data:** Evidence collected from experiment(s) (measurements or observations), the second part of a conclusion.
 - Example: the average weight of a dog is 14 kg and the average weight of a cat is 5 kg.

2. Directions: Circle CLAIM if the statement is a claim. Circle DATA if the statement is data. Circle NEITHER if the statement is neither claim nor data.

a) McDonalds served 100 customers and Taco Bell served 75	Claim	<input checked="" type="radio"/> Data	Neither
b) blue is the best color	Claim	Data	<input checked="" type="radio"/> Neither
c) butterflies that are larger than 15 cm are attracted to bright colors	<input checked="" type="radio"/> Claim	Data	Neither
d) ice has been observed to float on water	Claim	<input checked="" type="radio"/> Data	Neither
e) more people buy pizza than hamburgers	<input checked="" type="radio"/> Claim	Data	Neither
f) the average male blue whale weighs 100 tons, while the average female blue whale weighs 135 tons	Claim	<input checked="" type="radio"/> Data	Neither

10

Below are the explanations and answers to part 2 letters a-f on page 10.

Letter a: McDonalds served 100 customers and Taco Bell served 75 customers

Data (Data Collected: counted number of people)

This is data because a scientist needed to count the number of people coming out of McDonalds and Taco Bell. This could be paired with a claim to make the conclusion: McDonalds serves more customers than Taco Bell because McDonalds served 100 customers and Taco Bell served 75 customers.

Letter b: blue is the best color

Neither-Opinion

This statement is neither data nor a claim because it is an opinion and scientists cannot measure/observe opinions. An opinion can never be supported by data, therefore it cannot be a claim.

Letter c: butterflies that are larger than 15 cm are attracted to bright colors

Claim

This is not an example of data, because this statement was not measured or observed. Instead it is a claim. Tell students that the numbers that are present in this claim are called descriptive numbers because they describe the experiment that would need to be run to test this claim. These are numbers that identify a group; therefore, they are not a collected measurement. Tell students that we need be able to distinguish between descriptive numbers and collected measurements. Scientists can go out and collect data to support this claim. One piece of data that scientists could collect to support this claim is to count the number of butterflies that land on bright colored paper compared to the number of butterflies that land on black or brown paper. This could be paired with data to make the conclusion: butterflies that are larger than 15 cm are attracted to bright colors because 20 large butterflies landed on the bright papers (neon green, bright yellow, and turquoise) and only 2 landed on the black or brown paper.

Letter d: ice has been observed to float on water

Data (Data Collected: observed ice and water)

This is an example of data because a scientist went out and observed ice and water and saw that the ice floated on water. This could be paired with a claim to make the conclusion: ice is less dense than water because ice has been observed to float on water.

Letter e: people buy more pizza than hamburgers

Claim

This is not an example of data, because this statement was not measured or observed. Instead it is a claim. Scientists can go out and collect data to support this claim. One piece of data that scientists could collect to support the claim is to count the number of people that buy pizza and hamburgers in one day. This could be paired with data to make the conclusion: people buy more pizza than hamburgers because out of 10 people at the restaurant 7 people bought pizza and only 3 people bought hamburgers.

Letter f: the average male blue whale weighs 100 tons, while the average female blue whale weighs 135 tons

Data (Data Collected: measured the weight of blue whales (female and male))

This is an example of data because a scientist went out and measured the weight of female and male blue whales. This could be paired with a claim to make the conclusion: female blue whales weigh more than male blue whales because the average male blue whale weighs 100 tons, while the average female weighs 135 tons.

Once part 2 is completed have the students turn to page 11 in their notebooks.

Tell the students that now they are going to practice matching claims with supporting data. Tell the class to read the statements carefully because not all of the claims will make a match. Instruct them to only draw lines between the claims that match up with data. Tell the students to work by themselves for the first couple of minutes (~2 minutes) and that afterwards we will go over the examples as a class.

Ask the class if anyone has identified a match. Read each suggestion claim/data match that is made by a student and then ask the rest of the class if they agree/disagree using thumbs up/thumbs down. If they disagree, ask a student to explain. Continue asking students if they are able to make any other connections until all possible matches are made.

SCIENTIFIC PRACTICES
Conclusions

3. Directions: Draw a line connecting claims with the correct data. If there is no data that supports the claim, do not draw a line.

Claims	because	Data
1. People read more from electronic devices than books		a. Sony TVs give off 20 lumens of light and Samsung TVs give off 10 lumens of light.
2. Sony TVs are brighter than Samsung TVs	/	b. when blue and red paint were mixed together the paint was observed to turn purple.
3. The color purple is made from blue and red	/	c. wind turbines produce 6,000 MW of energy and solar panels produce 5,000 MW of energy in California.
4. Wind turbines produce less energy than solar panels in California		d. the speed of light is measured to be $3 \times 10^8 \frac{m}{s}$.

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Below are correct matches that can be made from this activity

1. Sony TVs are brighter than Samsung TVs because Sony TVs give off 20 lumens of light and Samsung TVs give off 10 lumens of light.
This is a correct match because the data clearly supports the claim using numerical values to make a conclusion.
2. The color purple is made from blue and red because when blue and red paint were mixed together the paint was observed to turn purple.
This is a correct match because the data clearly supports the claim using an observation to make a conclusion.

Below are incorrect matches that can be made from this activity

3. Wind turbines produced less energy than solar panels in California because wind turbines produce 6,000 MW of energy and solar panels produce 5,000 MW of energy in California.
This is an incorrect match because the data does not support the claim. The claim says that less energy is produced in wind turbines, however, the data supports the opposite claim that solar panels produce less energy. Ask students, in order to make a conclusion do you think scientists can change the claim or the data. Students should realize that scientists can change their claims but they cannot change data. In addition, scientists must include all data when generating a claim.
4. More people read from electronic devices than books because the speed of light is measured to be $3 \times 10^8 \frac{m}{s}$.
This is an incorrect match because the data has nothing to do with and does not support the claim. Therefore, this is an incorrect conclusion.

Wrap-Up (2 minutes – Full Class – SciTrek Lead):

Tell students that on the next SciTrek visit they will analyze others data from the pen dye experiments to identify appropriate claims and data statements. They will then analyze their data to draw a conclusion. After they will design a second experiment.

Clean-Up:

Before you leave, have students attach their nametags to their notebooks and place them in the group box. Make sure that all of the liquids and dishes are in the bucket and the buckets lid is securely fastened. Bring all materials back to UCSB. In addition, put your lab coat back in your group box.

Day 4: Conclusion Activity/Conclusion/Question/Experimental Set-Up/Procedure**Schedule:**

- Introduction (SciTrek Lead) – 2 minutes
- Conclusion Activity (SciTrek Lead) – 25 minutes
- Conclusion (SciTrek Volunteers) – 5 minutes
- Question (SciTrek Volunteers) – 5 minutes
- Experimental Set-Up (SciTrek Volunteers) – 10 minutes
- Procedure (SciTrek Volunteers) – 10 minutes
- Wrap-Up (SciTrek Lead) – 3 minutes

Materials:**(3) Volunteer Boxes:**

- | | | |
|---|---|---------------------------------------|
| <input type="checkbox"/> Student nametags | <input type="checkbox"/> (3) Materials pages (group color/number indicated) | <input type="checkbox"/> Pencil |
| <input type="checkbox"/> Student notebooks | <input type="checkbox"/> Volunteer lab coat | <input type="checkbox"/> (2) Red pens |
| <input type="checkbox"/> Volunteer instructions | | <input type="checkbox"/> Notepad |

Lead Box:

- | | | |
|--|--|---------------------------------------|
| <input type="checkbox"/> (3) Extra student notebooks | <input type="checkbox"/> (3) Materials pages | <input type="checkbox"/> (2) Pencils |
| <input type="checkbox"/> Time card | <input type="checkbox"/> Lead lab coat | <input type="checkbox"/> (2) Red pens |
| <input type="checkbox"/> Lead instructions | | <input type="checkbox"/> Notepad |

SciTrek Notebook Pages Used with Students:

SCIENTIFIC PRACTICES
Conclusions

Variables:		Trial A	Trial B
Time:	5 min	→	→
Liquid Type:	Water	→	→
Liquid Amount:	2 mL	→	→
Paper Type:	Original	→	→
Pen Color:	Black	→	→
Pen Type:	Mr. Sketch	→	Crayola
Initial Dot Height:	2 cm	→	→
Data:		Trial A	Trial B
Final Observations/Measurements:	Liquid Height:	4 cm	3 cm
	Smear Height:	5 cm	2 cm
	Other:		

Directions: Identify the following statements as either CLAIM, DATA, or OPINION. Once you have identified the statement type, write a C, D, or O in the box. Then look at the data table to determine if the statement is correct or incorrect. Circle CLAIM if the statement is an appropriate claim that can be made from the given results. Circle DATA if the statement is data that is consistent with the results. Circle OPINION if the statement is an opinion. Circle INCORRECT if the statement does not agree with the data table or has not been tested.

- the paper type affects the height the liquid travels up the paper C Claim Data Opinion INCORRECT
- black pen types are made up of different dye colors C CLAIM Data Opinion INCORRECT
- when a black dot sits in water for 5 min, different pen types give different smear heights C CLAIM Data Opinion INCORRECT
- black pens are made of pretty colored dyes O Claim Data OPINION INCORRECT

What data can be used to support claim b above? black Mr. Sketch was observed to contain green, blue, and red dyes while black crayola was observed to contain yellow, blue, and red dyes.

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SCIENTIFIC PRACTICES
Conclusions

Variables:		Trial A	Trial B
Time:	5 min	→	4 min
Liquid Type:	Water	→	Soap
Liquid Amount:	2 mL	→	→
Paper Type:	Original	→	→
Pen Color:	Black	→	Red
Pen Type:	Mr. Sketch	→	→
Initial Dot Height:	2 cm	→	→
Data:		Trial A	Trial B
Final Observations/Measurements:	Liquid Height:	4 cm	2 cm
	Smear Height:	3 cm	1.5 cm
	Other:		

- the stronger the pen odor the larger the smear height C Claim Data Opinion INCORRECT
- the black pen had a smear height of 3 cm and the red pen had a smear height of 1.5 cm D Claim Data Opinion INCORRECT
- red pens are made from green dye C Claim Data Opinion INCORRECT
- the thicker the liquid the shorter the smear height C Claim Data Opinion INCORRECT

If no claim can be made from the data state why not. NO CLAIM CAN BE MADE BECAUSE THERE IS MORE THAN ONE CHANGING VARIABLE.

If no claim can be made from the results, can you make a conclusion?
 YES NO

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SCIENTIFIC PRACTICES
Conclusions

Decide if a claim/conclusion can be made for each of the following results tables.

Table A

Variables:		Trial A	Trial B
Time:	3 min	→	2 min
Pen Type:	Crayola	→	Mr. Sketch
Pen Color:	Black	→	→
Paper Type:	Original	→	→
Liquid:	Water	→	→
Liquid Amount:	2 mL	→	4 mL
Data:		Trial A	Trial B
Smear Height:	5 cm	→	4 cm

Can this person make a claim/conclusion? NO

Table B

Variables:		Trial A	Trial B
Time:	5 min	→	→
Pen Type:	Mr. Sketch	→	→
Pen Color:	Blue	→	→
Paper Type:	Manila	→	Original
Liquid:	Etanage	→	→
Liquid Amount:	5 mL	→	→
Data:		Trial A	Trial B
Smear Height:	2.2 cm	→	4 cm

Can this person make a claim/conclusion? YES

Table C

Variables:		Trial A	Trial B
Time:	5 min	→	→
Pen Type:	Blue Art	→	Expo
Pen Color:	Black	→	→
Paper Type:	Original	→	→
Liquid:	Soap	→	→
Liquid Amount:	2 mL	→	→
Data:		Trial A	Trial B
Smear Height:	3 cm	→	3 cm

Can this person make a claim/conclusion? YES

Graph D

Changing Variable(s)	Trial	Smear Height (cm)
Pen Color: Red, Liquid Amount: 2 mL	B	3 cm
Pen Color: Blue, Liquid Amount: 4 mL	A	4.5 cm
Pen Color: Black, Liquid Amount: 5 mL	C	6 cm
Pen Color: Green, Liquid Amount: 3 mL	D	8 cm

Can this person make a claim/conclusion? NO

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Making a Conclusion from Your Data

How many changing variables did you have in your experiment? 2

Can you make a conclusion from your data? YES NO

IF NO

Why? I can not make a conclusion because my experiment had too many changing variables.

IF YES (Make sure to look back at your question (page 5) before you generate your conclusion.)

CONCLUSION

We can conclude _____ claim

because _____ data (measurement/observation)

SciTrek Member Approval [Signature]

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Changing Variables

For your second experiment decide which variable(s) (max three) that you would like to test.

Changing Variable 1: liquid type Changing Variable 2: _____
 Changing Variable 3: _____

QUESTION

Question our group will investigate:

- If we change the liquid type _____
insert each changing variable

what will happen to the height and color of the smear _____
insert what you are measuring/observing

SciTrek Member Approval Sg

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EXPERIMENTAL SET-UP

Determine the values of your changing variable(s) (ex: pen color) from the materials page and write the values (ex: trial I: blue) for your four trials under each strip of paper.

Changing Variable(s):

1) liquid type : RA water soap vinegar

2) _____

3) _____

Controls (variables you will hold constant):
 *Write at least four variables that you will hold constant and indicate the specific value you will use in all your trials.

Test tube / Original		<u>pen type / crayola</u>	
<u>time / 5 min.</u>		<u>liquid amount / 3ml</u>	
<u>paper type / original</u>		_____	

SciTrek Member Approval Sg

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PROCEDURE

Procedure Note:
 Make sure to include all values of your changing variable(s) in the procedure. (Example: for a group that decided to change pen color one step would be: put colored dot with a Mr. Sketch pen on original paper at 2 cm E) red, F) blue, G) green, and H) yellow.)

1. Fill four test tubes with 3ml of the following:
 E) RA F) water G) Soap H) vinegar _____
2. Put a dot on original paper with a black Crayola marker 2cm from the bottom _____
3. Put papers in test tubes _____
4. Time for 5 minutes _____
5. Remove papers from test tubes _____
6. Record length of the smear and observe smear color _____
7. _____

SciTrek Member Approval Sg

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Set-Up:

SciTrek Lead:

If the classroom has a document camera, ask the teacher to use it for the conclusion activity (pages 12- 14). If the classroom does not have a document camera, then tape example poster-size notebook pages to the front board. Have students sit in their regular classroom seats when completing the conclusion activity.

SciTrek Volunteer:

Pass out the SciTrek notebooks/nametags to the students in their regular seats. The first part of the activity will be done in their regular classroom seats, then students will break up into their small groups.

Introduction (2 minutes – Full Class – SciTrek Lead):

Have the students sit down in their regular classroom seats. While you are doing the introduction have the SciTrek volunteers hand out the SciTrek notebooks/nametags to the students.

Inform students that today they are going to analyze other's data to determine which claims and data are appropriate for a given set of results. Then, they will analyze their own data to see if they can make a claim/conclusion. They will then have the opportunity to design a second experiment or redesign their first experiment, which will be carried out during the next SciTrek visit.

Conclusion Activity (25 minutes – Full Class – SciTrek Lead):

Tell the students to turn to page 12 in their notebooks. Place a blank notebook under the document camera and open to page 12.

Ask students what types of statements are needed to make a conclusion. Students should tell you that a conclusion is made from a claim and a supporting data statement. Ask students for the definition of a claim. Students should remind you that a claim is the explanation of your data, a statement that can be verified by testing. Ask students what types of things can be used for data. Students should remind you that data is either measurements or observations. Ask them if all statements have to be either claim or data. Students should say no; there are many other type of statements, one being opinions that are neither claim nor data statements. Tell the students that we are now going to look over a list of statements about a given data set and then decide if each statement is an example of a claim, data, or an opinion. In addition since we have the results table, we can also check to see if the statement is incorrect; the claim is incorrect if it does not agree with the data or if the statement was not tested.

Tell the students to look over the data table. Ask a student to explain the experiment that was conducted. Make sure that they identify changing variables and controls. For example one explanation might be, "In this experiment, a black dot was drawn on two strips of the original paper with two different pen types. The strips were then put in water for 5 minutes. The changing variable was the pen type and the controls were the time, liquid type, liquid amount, paper type, pen color, and initial dot height."

Tell students we are now going to examine several statements about the data table that we just examined and classify each statement as (claim, data, opinion, or incorrect). The first thing that we will do is read

the statement and decide if the statements is claim, data, or opinion. If the statement is a possible claim they should write a “c” on the line by the statement, if the statement is possible data they should write a “d” on the line by the statement, if the statement is an opinion they should write an “o” on the line by the statement. Once they have identified the type of statement if the statement is a claim or data they will need to check to see if it is correct. For claims, the statement will be incorrect if it does not agree with the data or has not been tested. For data, the statement will be incorrect if it does not agree with the results.

Letter a: the paper type affects the height the liquid travels up the paper

Incorrect (Claim/Variable Held Constant)

Students should identify this statement as a possible claim and write “c” next to the statement. This is a possible claim because the statement was not measured or observed and it is not an opinion. Ask the students if the scientists tested this claim and if the claim is consistent with the data. Students should see that the claim is made about the paper type; however, in the experiment this was a variable that was held constant. Since no data was collected on the effects of changing the paper type there is no data that could support this claim. Therefore, this claim is inappropriate for the results table and students should circle incorrect.

Letter b: black pen types are made up of different dye colors

Correct

Students should identify this statement as a possible claim and write “c” next to the statement. This is a possible claim because the statement was not measured or observed and it is not an opinion. Ask the students if the scientists tested this claim and if the claim is consistent with the data. Students should see that the claim is made about the pen type. Pen type was the changing variable in the experiment. In addition, the different pen types had different dye colors. Therefore, this claim is consistent with the results table and students should circle claim.

Letter c: when a black dot sits in water for 5 min, different pen types give different smear heights

Claim

Students should identify this statement as a possible claim and write “c” next to the statement. This is a possible claim because the statement was not measured or observed and it is not an opinion. The number that is present in this claim (5 min) is called a descriptive number because it describes the experiment we would need to test this claim; therefore, the number is not a collected measurement. Ask the students if the scientists tested this claim and if the claim is consistent with the data. Students should see that the claim is made about the pen type. Pen type was one of the changing variables in the experiment. In addition, the different pen types have different smear heights. Therefore, this claim is consistent with the results table and student should circle claim.

Letter d: black pens are made from pretty colored dyes

Opinion

Students should identify this statement as an opinion and write “o” next to the statement. This is an opinion because the statement was not measured or observed and it cannot be tested. The statement cannot be tested because different people will think that different colors are pretty, therefore, students should circle opinion.

Tell students we are now going to determine the data to support claim b. Have students fill in the claim letter in the following sentence frame, “What data can be used to support claim b above? Read claim b aloud (black pens are made up of different dye colors) and ask the students what data, from the results

table, can be used to support this claim. They should respond that black Mr. Sketch was observed to contain green, blue, and red dyes while black Crayola contains yellow, blue and red dyes. Ask students how people know that the statement we just generated was data. They should reply that it contains an observation. Have students record this in their notebook. If there is time you can have the students determine the data to support claim c: the smear height for Mr. Sketch was 3 cm while the smear height for Crayola was 2 cm. This is data because it contains a measurement.

SCIENTIFIC PRACTICES
Conclusion

Variables:		Trial A	Trial B
Time:		5 min	
Liquid Type:		Water	
Liquid Amount:		2 ml	
Paper Type:		Original	
Pen Color:		Black	
Pen Type:		Mr. Sketch	Crayola
Initial Dot Height:		2 cm	
Data:		Trial A	Trial B
Final Observations/Measurements:	Liquid Height:	4 cm	3 cm
	Smear Height:	5 cm	2 cm
	Other:		

Directions: Identify the following statements as either CLAIM, DATA, or OPINION. Once you have identified the statement type, write a C, D, or O as the box. Then look at the data table to determine if the statement is correct or incorrect. Circle CLAIM if the statement is an appropriate claim that can be made from the given results. Circle DATA if the statement is data that is consistent with the results. Circle OPINION if the statement is an opinion. Circle INCORRECT if the statement does not agree with the data table or has not been tested.

- the paper type affects the height the liquid travels up the paper. C Data Opinion INCORRECT
- black pen types are made up of different dye colors. C CLAIM Data Opinion INCORRECT
- when a black dot sits in water for 5 min, different pen types give different smear heights. C CLAIM Data Opinion INCORRECT
- black pens are made of pretty colored dyes. O Claim Data OPINION INCORRECT

What data can be used to support claim b above? black Mr. Sketch was observed to contain green, blue, and red dyes while black Crayola was observed to contain yellow, blue, and red dyes.

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Have students turn to page 13 in their notebooks. Turn the example notebook to page 13.

Tell the students to look over the data table. Ask a student to explain the experiment that was conducted. Make sure that they identify changing variables and controls. For example one explanation might be, "In this experiment, a different color dot was draw on two strips of the original paper. The papers were in two different liquids for two different times. The changing variables were time, liquid type, and pen color and the controls were liquid amount, paper type, pen type, and initial dot height." Tell the students that we are now going to go through the same process that we went through with the last data set, identifying each of the given statements as claim, data, opinion, or incorrect.

Letter a: the stronger the pen odor the larger the smear height
Incorrect (Claim/No Data Gathered)

Students should identify this statement as a possible claim and write "c" next to the statement. This is a possible claim because the statement was not measured or observed and it is not an opinion. Ask the students if the scientists tested this claim and if the claim is consistent with the data. Students should see that the claim is made about the odor of the pens, however, this variable that was not measured in this experiment. Since no data was collected on the effects of

odor on smear height there is no data that could support this claim. Therefore, this claim is inappropriate for the results table and students should circle incorrect.

Letter b: the black pen had a smear height of 3 cm and the red pen had a smear height of 1.5 cm
Data

Students should identify this statement as possibly data and write “d” next to the statement. The statement is possibly data because it contains a measurement. Ask the students if the statement is consistent with the results. Students should see that the results agree with the statement. Therefore, this data is consistent with the results table and students should circle data.

Letter c: red pens are made from green dye
Incorrect (Claim/Inconsistent with Data)

Students should identify this statement as a possible claim and write “c” next to the statement. This is a possible claim because the statement was not measured or observed and it is not an opinion. Ask the students if the scientists tested this claim and if the claim is consistent with the data. Students should see that the claim is made about the paper color. Pen color was one of the changing variables in the experiment. However, the red pen only has red, pink, and orange and not green. Therefore, the claim is inconsistent with the results table and students should circle incorrect.

Letter d: the thicker the liquid the shorter the smear height
Incorrect (Claim/More than One Changing Variable)

Students should identify this statement as a possible claim and write “c” next to the statement. This is a possible claim because the statement was not measured or observed and it is not an opinion. Ask the students if the scientists tested this claim and if the claim is consistent with the data. Students should see that the claim is made about the liquid type. Liquid type was one of the changing variables in the experiment, however, the time and pen color were also changing variables and it is impossible to tell which of the three variables affected the smear height. Therefore, the claim is inappropriate for the results table and students should circle incorrect.

Ask the students why no claims can be made from the data. They should say that because there is more than one changing variable you cannot tell which variable affected the results or how/why these changing variables affected the smear. Record this answer on the example notebook and have students copy this into their notebooks.

Ask the students if they think they would be able to make a conclusion when a claim cannot be made from the data. Since a conclusion is defined as a claim supported data, you would need both a claim and data to make a conclusion. Since no claims can be made, a conclusion cannot be made either. Have students check the “no” box in their notebook.

SCIENTIFIC PRACTICES
Conclusions

Variables:		Trial A	Trial B
Time:		5 min	4 min
Liquid Type:		Water	Soap
Liquid Amount:		2 ml	
Paper Type:		Original	
Pen Color:		Black	Red
Pen Type:		Mr. Sketch	
Initial Dot Height:		2 cm	
Data:		Trial A	Trial B
Final Observations/Measurements:	Liquid Height:	4 cm	2 cm
	Smear Height:	3 cm	1.5 cm
	Other:		

a) the stronger the pen odor the larger the smear height C Claim Data Opinion Incorrect

b) the black pen had a smear height of 3 cm and the red pen had a smear height of 1.5 cm D Claim Data Opinion Incorrect

c) red pens are made from green dye C Claim Data Opinion Incorrect

d) the thicker the liquid the shorter the smear height C Claim Data Opinion Incorrect

If no claim can be made from the data state why not. NO claim can be made because there is more than one changing variable.

If no claim can be made from the results, can you make a conclusion?
 YES NO

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Ask the students what they learned about conclusions from this activity. Make sure by the end of the conversation that students understand that in order to draw a conclusion they must only have one changing variable.

Tell students they are now going to look at three results tables and one bar graph and determine which data sets would allow them to make a claim/conclusion. Tell them to look at A- D, and underline the changing variable(s) for each table. They should then decide if a claim/conclusion can be made from the results. Give students ~1 minute to try the activity by themselves before discussing the pictures as a class.

SCIENTIFIC PRACTICES
Conclusions

Decide if a claim/conclusion can be made for each of the following results tables.

Table A

Variables	Trial A	Trial B
Time	3 min	2 min
Pen Type	Gelball	Mr. Sketch
Pen Color	Black	
Paper Type	Original	
Liquid	Water	
Liquid Amount	2 mL	4 mL
Date:	Trial A	Trial B
Smear Height	3 cm	4 cm

Can this person make a claim/conclusion? NO

Table B

Variables	Trial A	Trial B
Time	5 min	
Pen Type	Mr. Sketch	
Pen Color	Blue	
Paper Type	Newsprint	Original
Liquid	Windex	
Liquid Amount	3 mL	
Date:	Trial A	Trial B
Smear Height	2.2 cm	4 cm

Can this person make a claim/conclusion? YES

Table C

Variables	Trial A	Trial B
Time	3 min	
Pen Type	Sharpie	Edge
Pen Color	Black	
Paper Type	Original	
Liquid	Soup	
Liquid Amount	2 mL	
Date:	Trial A	Trial B
Smear Height	3 cm	3 cm

Can this person make a claim/conclusion? YES

Graph D

Trial	Smear Height (cm)
A	2.2
B	4.0
C	3.0
D	4.0

Can this person make a claim/conclusion? NO

Ask students the following questions:

Table A

What is/are the changing variable(s)?

time, pen type, and liquid amount

How many changing variables are there?

three

Can a conclusion/claim be made from this data?

no

Why not?

This experiment had three changing variables and conclusions/claims can only be made when there is one changing variable.

Table B

What is/are the changing variable(s)?

paper type

How many changing variables are there?

one

Can a conclusion/claim be made from this data?

yes

Did the paper type affect the smear height?

yes

Table C

- What is/are the changing variable(s)?
pen type
- How many changing variables are there?
one
- Can a conclusion/claim be made from this data?
yes
- Did the pen type effect the smear height?
no

Graph D

- What is/are the changing variable(s)?
pen color , and liquid amount
- How many changing variables are there?
two
- Can a conclusion/claim be made from this data?
no
- Why not?
They had two changing variables and conclusions/claims can only be made when there is one changing variable.

Tell students that tables (such as tables A-C) and graphs (such as graph D) represent two different ways of displaying results from an experiment. Ask the students what the advantages and disadvantages are for tables versus graphs. Students should realize that viewing at the data in table form yields a complete idea about what experiment was conducted and which controls were used; however, it is harder to see patterns in the data. Alternatively, viewing the data in graph form allows patterns or trends to be viewed but does not display the controls the experimenter used.

Inform students that scientists often graph their own data so that they are better able to understand their results and to look for patterns and trends that can best explain their data. Point out that in Graph D Trial B is graphed before Trial A. Ask students why they think the data was graphed in this way. Students should respond that it was graphed this way to show a pattern. Tell students that when they make their own graphs they can graph the trials in any order so that they might be able to better see trends or patterns in their results.

Tell students that they will now analyze their data to see if they can make a conclusion. Notify them that the group that comes up with the best conclusion, or claim supported by data, will get a prize on the last day of the module. Remind them that it is okay if they cannot draw a conclusion from their first experiment because they will have the opportunity to run another experiment in which they should only have one changing variable so that they will be able to draw a conclusion.

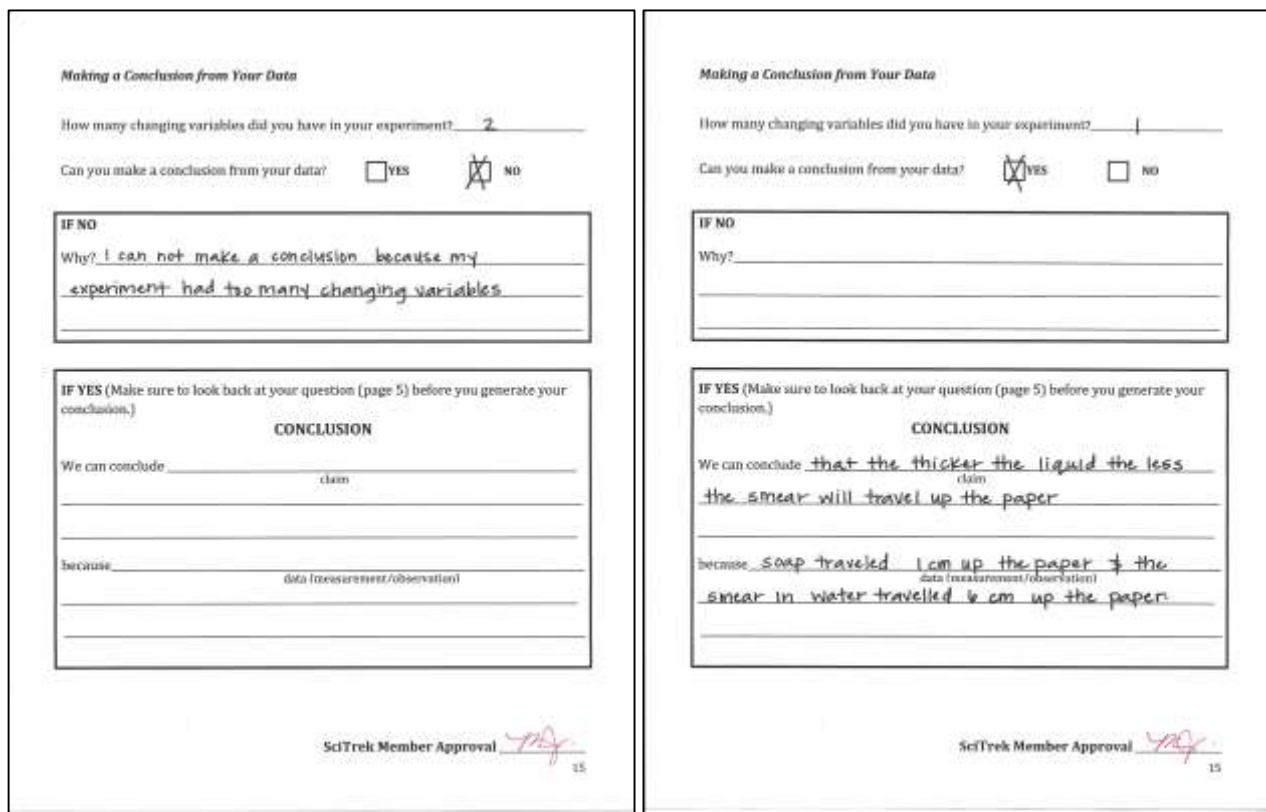
Inform students that once they have decided if they can/cannot make a conclusion they will either state why they cannot make a conclusion or use their results to make a conclusion. Once this is complete they can move on to designing their new experiment. Tell them that they are going to give poster presentations at the end of the module and the presentations will be more interesting if there are multiple changing variables that have been tested. In addition if a wide range of variables are chosen, the class question (What variables affect the smear?) will be more likely to be answered. Therefore, they should try to explore a changing variable that they think no one else in the class is investigating.

Tell students they will now start working with their group to analyze their old experiment and start their new experiment.

Conclusion (5 minutes – Small Groups – SciTrek Volunteers):

Help groups fill out page 15 of their notebook. If the group has more than one changing variable they will not be able to draw a conclusion. An example of when the students cannot make a conclusion is seen below on the left.

If the group has only one changing variable they will be able to make a conclusion. Make sure that the students have both a claim and supporting data and that these statements are in the appropriately labeled sections. Even if their conclusion is contrary to what you think, have students make a claim based solely on their data. Conclusions are still valid, and important, if they show that the changing variable tested did not affect the smear. Remind students that they should include either a measurement or an observation into their data statement so that other scientists will know that they collected the data. If you think that their data is flawed it is okay to ask them what they think went wrong and encourage them to repeat their experiment if it only had one changing variable. An example of when the students can make a conclusion is seen below on the right.



Left Page (Cannot make a conclusion):

Making a Conclusion from Your Data

How many changing variables did you have in your experiment? 2

Can you make a conclusion from your data? YES NO

IF NO

Why? I can not make a conclusion because my experiment had too many changing variables

IF YES (Make sure to look back at your question (page 5) before you generate your conclusion.)

CONCLUSION

We can conclude _____ claim

because _____ data (measurement/observation)

SciTrek Member Approval [Signature] 15

Right Page (Can make a conclusion):

Making a Conclusion from Your Data

How many changing variables did you have in your experiment? 1

Can you make a conclusion from your data? YES NO

IF NO

Why? _____

IF YES (Make sure to look back at your question (page 5) before you generate your conclusion.)

CONCLUSION

We can conclude that the thicker the liquid the less claim
the smear will travel up the paper

because SOSP traveled 1 cm up the paper & the data (measurement/observation)
smear in water travelled 6 cm up the paper

SciTrek Member Approval [Signature] 15

Question (5 minutes – Small Groups – SciTrek Volunteers):

Have students determine what their changing variable will be for their second experiment. Make sure each group has only one changing variable so they will be able to make a claim/conclusion after their experiment. If possible, encourage your small groups to each have a different changing variable. After

groups have decided on their changing variable have them fill out their question. When you sign off on their question you can give them a materials page with their group color and number designated in the upper right hand corner.

Select one group member to read their question they are going to investigate to the class during the wrap-up discussion.

Changing Variables

For your second experiment decide which variable(s) (max three) that you would like to test.

Changing Variable 1: liquid type Changing Variable 2: _____


Changing Variable 3: _____

QUESTION

Question our group will investigate:

• If we change the liquid type
insert each changing variable

what will happen to the height and color of the smear
insert what you are measuring/observing

SciTrek Member Approval: 

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Experimental Set-Up (10 minutes – Small Groups – SciTrek Volunteers):

Have groups determine the values for their changing variable and record them in their experimental set-up on page 17. Then encourage students to determine what their controls will be and have them record these constants on the left side of the line in their experimental set-up on page 17. After have students determine the values of their controls using the materials page as a reference. Ask students to justify the values that they have chosen for their controls and have them tell you if these values will make it easier or harder to answer their question. In addition, collect the materials page and make sure that it is filled out correctly. It is essential for students to do this so that they can start their experiments during the next class. An example of a notebook and materials page is seen below.

EXPERIMENTAL SET-UP

Determine the values of your changing variable(s) (ex: pen color) from the materials page and write the values (ex: trial I: blue) for your four trials under each strip of paper.

Changing Variable(s):

1) liquid type: RA water soap vinegar

2) _____

3) _____

Controls (variables you will hold constant):
 *Write at least four variables that you will hold constant and indicate the specific value you will use in all your trials.

Test tube / Original: pen type / crayola
 time / 5 min. liquid amount / 3mL
 paper type / original

SciTrek Member Approval: Sg

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MATERIALS PAGE

Color: orange
Group Number: 3

You will only have access to the following materials. Circle the materials that you need for your experiment.

General Materials:
 2 test tubes with corks (circled) test tube rack (circled) timer (circled)
 2 rulers (circled) 2 droppers (circled) 2 graduated cylinders (circled)

Types of Liquid:
 rubbing alcohol (RA) (circled) soap (circled) vinegar (circled)
 water (circled)

Types of Paper: You may only have 4 strips of paper (each paper that you circle put the number of strips that you need on the line. All of the numbers should add up to 4.)
 original paper (this is the paper that was used on the first day in the group experiment) 4
 coffee filter _____ construction paper _____ copy paper _____
 graph paper _____ newspaper _____ paper towel _____
 copy paper pink _____ copy paper yellow _____ copy paper green _____
 copy paper purple _____ copy paper blue _____

Types of Pens/Colors:

Mr. Sketch (original)	red	orange	yellow	light green
	dark green	light blue	dark blue	purple
	light pink	dark pink	black (original)	brown
Crayola	red	orange	yellow	green
	blue	purple	black	brown
Epa	red	orange	yellow	green
	blue	purple	black	brown
Sharpie	red	orange	yellow	green
	blue	purple	black	brown
Base Art	red	orange	yellow	green
	blue	purple	black	brown

Types of Black Pen:
 Bic _____ Crayola (circled) Dry Erase
 Epa _____ Mr. Sketch (circled) Paper Mate
 Base Art _____ Sharpie (circled)

Procedure (10 minutes – Small Groups – SciTrek Volunteers):

After each small group has filled out their experimental set-up they can start on their procedure (page 18). Keep procedures as brief as possible with still conveying the pertinent information about the experiment. Make sure that you have students include all changing variables values in the procedure. For example, if pen color was one of the changing variables, “put colored dot with Mr. Sketch pen on original paper at 2 cm E) red, F) blue, G) green, and H) yellow” would be one of the procedural steps. If they are having problems with their procedure they should look back at their initial procedure on page 7 of their notebook. If they are still having trouble, you can have these groups dictate each step while you transcribe them onto a notepad found in the group box. Give this sheet to the students to copy into their notebooks. Once the students have their procedure written in their notebooks they should raise their hands to get approval by their SciTrek group volunteer. An example procedure can be seen below.

If groups have extra time you can have them fill out their results table.

PROCEDURE

Procedure Note:
Make sure to include all values of your changing variable(s) in the procedure. (Example, for a group that decided to change pen color one step would be: put colored dot with a Mr. Sketch pen on original paper at 2 cm E) red, F) blue, G) green, and H) yellow.)

1. Fill four test tubes with 3ml of the following:
E) RA F) water G) Soap H) vinegar
2. Put a dot on original paper with a black Crayola marker 2cm from the bottom
3. Put papers in test tubes
4. Time for 5 minutes
5. Remove papers from test tubes
6. Record length of the smear and observe smear color
- 7.

SciTrek Member Approval: SG

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Wrap-Up (3 minutes – Full Class – SciTrek Lead):

Have one group from each of the large groups read the question that they are going to investigate. Tell the students that you are really excited to see their experiments and they will get a chance to perform these experiments the next SciTrek visit.

Clean-Up:

Before you leave, have students attach their nametags to their notebooks and place them in the group box. Place the materials pages on top of the notebooks in your group box. Bring all materials back to UCSB. In addition, put your lab coat back in your group box.

Day 5: Experiment/Conclusion

Schedule:

- Introduction (SciTrek Lead) – 20 minutes
- Results Table (SciTrek Volunteers) – 5 minutes
- Experiment (SciTrek Volunteers) – 20 minutes
- Bar Graph (SciTrek Volunteers) – 5 minutes
- Conclusion (SciTrek Volunteers) – 8 minutes
- Wrap-Up (SciTrek Lead) – 2 minutes

Materials:

(3) Volunteer Boxes:

- | | | |
|---|--|--|
| <input type="checkbox"/> Student nametags | <input type="checkbox"/> (2) Red pens | <input type="checkbox"/> Vinegar |
| <input type="checkbox"/> Student notebooks | <input type="checkbox"/> Notepad | <input type="checkbox"/> Water |
| <input type="checkbox"/> Volunteer instructions | <input type="checkbox"/> (3) Test tube stands | <input type="checkbox"/> Soap |
| <input type="checkbox"/> Volunteer lab coat | <input type="checkbox"/> (12) Small cups (labeled with liquid types) | <input type="checkbox"/> Rubbing alcohol |
| <input type="checkbox"/> (3) Pencils | | <input type="checkbox"/> Paper towels |
| | | <input type="checkbox"/> (6) Rulers |

(3) Ziploc Bags labeled group 1, 2, and 3 each with the following:

- | | | |
|--|--|--------------------------------------|
| <input type="checkbox"/> (2) 10 ml Graduated cylinders | <input type="checkbox"/> (2) Droppers | <input type="checkbox"/> Paper towel |
| <input type="checkbox"/> (4) Corks | <input type="checkbox"/> Requested pens | <input type="checkbox"/> Timer |
| | <input type="checkbox"/> Requested strips of paper | |

Other Supplies:

- | | |
|--|--|
| <input type="checkbox"/> Box of test tubes | <input type="checkbox"/> Bucket with lid |
|--|--|

Lead Box:

- | | | |
|--|---|---|
| <input type="checkbox"/> (3) Student notebooks | <input type="checkbox"/> (4) 10 ml Graduated cylinders | <input type="checkbox"/> (5) Bags of different colored markers (Mr. Sketch, Crayola, Expo, Sharpie, and Rose Art) |
| <input type="checkbox"/> Lead instructions | <input type="checkbox"/> Bag of chromatography paper (minimum 30) | <input type="checkbox"/> Vinegar |
| <input type="checkbox"/> Time card | <input type="checkbox"/> Bag with the 11 different kinds of papers (minimum 20 each) | <input type="checkbox"/> Water |
| <input type="checkbox"/> Lead lab coat | <input type="checkbox"/> (8) Droppers | <input type="checkbox"/> Soap |
| <input type="checkbox"/> (2) Pencils | <input type="checkbox"/> Bag of 8 different black pens (Bic, Crayola, Dry Erase, Expo, Mr. Sketch, Paper Mate, Rose Art, and Sharpie) | <input type="checkbox"/> Rubbing alcohol |
| <input type="checkbox"/> (2) Red pens | | <input type="checkbox"/> (4) Rulers |
| <input type="checkbox"/> Notepad | | <input type="checkbox"/> (2) Timers |
| <input type="checkbox"/> (2) Test tube stands | | <input type="checkbox"/> Paper towels |
| <input type="checkbox"/> (10) Corks | | |
| <input type="checkbox"/> (12) Small cups (labeled with liquid types) | | |

SciTrek Notebook Pages Used with Students:

RESULTS Table

Fill out the chart for each of your trials. If one of the variables remains constant for all trials write the value in trial E and then draw a line through each box indicating that this variable is control.

Variables:	Trial E	Trial F	Trial G	Trial H	
Test Tube:	Original	—————→			
Time:	5 min	—————→			
Liquid Type:	RA	water	soap	vinegar	
Liquid Amount:	3 mL	—————→			
Paper Type:	original	—————→			
Pen Color:	black	—————→			
Pen Type:	crayola	—————→			
Initial Dot Height:	2 cm	—————→			
<i>How to Use</i>					
Predictions:	Trial E	Trial F	Trial G	Trial H	
Put a "T" in the trial that will give the tallest smear and an "S" in the trial that will give the shortest smear.		T	S		
Data:	Trial E	Trial F	Trial G	Trial H	
Final Observations/Measurements	Liquid Height:	6.5 cm	8 cm	3 cm	5 cm
	Smear Height:	4.5 cm	6 cm	1 cm	3 cm
	Other:	blue, green, yellow	yellow, blue, green	yellow, blue, green	yellow, blue, green

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SCIENTIFIC PRACTICES Conclusions

Variables:	Trial A	Trial B	Trial C	Trial D
Test Tube:	Original	—————→		
Time:	5 min	—————→		
Liquid Type:	Water	—————→		
Liquid Amount:	2 ml	—————→		
Paper Type:	Original	—————→		
Pen Color:	Black	—————→		
Pen Type:	Mr. Sketch	—————→		
Initial Dot Height:	2 cm	—————→		
Paper Height:	20 cm	15 cm	10 cm	5 cm
Data:	Trial A	Trial B	Trial C	Trial D
Liquid Height:	11 cm	11 cm	10 cm	5 cm
Smear Height:	9 cm	9 cm	8 cm	3 cm
Other:	blue, orange, red	blue, orange, red	blue, orange, red	blue, orange, red

Write a conclusion from the results above:

We can conclude the liquid will reach the top of the paper if the paper is 11 cm or shorter ^{claim}

because when the paper height was 5 cm and 10 cm the liquid level was 5 cm and 10 cm and when the paper height was 15 cm and 20 cm the liquid level was only 11 cm. ^{data}

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RESULTS Graph

Record what you measured for each trial (example: smear height) on the y-axis (vertical axis) and each of your changing variable(s) values (example: red/water) on the x-axis. There may be blank lines on your x-axis. Determine an appropriate scale and record your data from each of your trials on the chart.

Trial	Smear Height (cm)
E	4.5
F	6
G	1
H	3

Changing Variable(s):
 liquid type: RA water soap vinegar

Compare Predictions and Actual Data
 Look at your predicted and actual smear heights (page 19). Were you surprised by your results why or why not? Discuss with group differences between your actual and predicted data.

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What was your question? (page 16)

- If we change the liquid type ^{independent changing variable} what will happen to the height and color of the smear? ^{infer what you are measuring/observing}

Explain how your changing variable did or did not affect your results. (Example: the shorter the paper the shorter the smear height.) The thicker the liquid type the smaller the height of the smear

this is your claim

Can you test the above statement? (Is the above statement a claim?)

YES NO (If you checked this box go back and revise your statement so that it can be tested which will make it a claim.)

What data do you have to support your claim? (Remember to include your measurements or observations.) the thick liquid, soap, had a smear height of 1 cm while the thin liquid, water, had a smear height of 6 cm.

this is your data

How do you know the statement above is data? I know this is data because it contains a measurement.

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Set-Up:

SciTrek Lead:

If the classroom has a document camera, ask the teacher to use it for the conclusion example (page 20). If the classroom does not have a document camera, then tape up the poster-size notebook pages on the front board.

SciTrek Volunteer:

While the SciTrek lead is giving the introduction pass out the SciTrek notebooks/nametags to the students in your group. Once the notebooks/nametags are passed out place the test tubes in the test tube rack and pour all of the liquids that your groups need into the small cups. Have all supplies ready so that you can set them out as soon as your groups are ready to start. After the introduction, when students come to your group make sure that students with the same number are sitting next to each other.

Introduction (20 minutes – Full Class – SciTrek Lead):

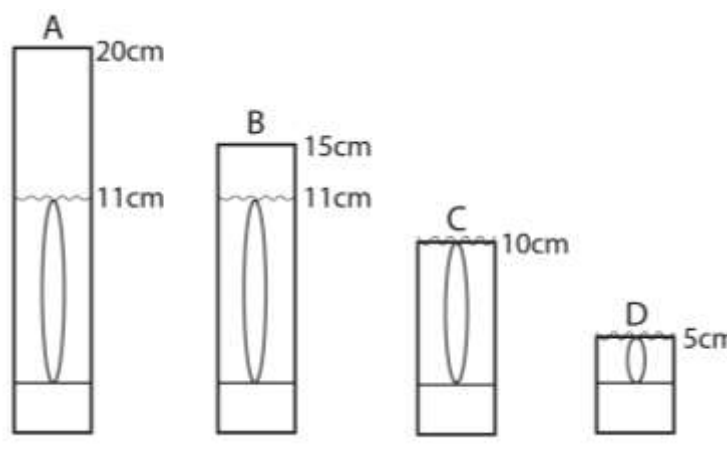
Have the students sit in their regular classroom seats. While you do the introduction have the SciTrek volunteers hand out the notebooks/nametags to the students and then set-up for the experiment.

Ask the class what the class question is that we have been investigating. Students should tell you, “What variables affect the smear?” Tell the students that today they are going to perform their second experiment to help answer the class question. Once the experiment is complete they will analyze their data and determine what conclusions can be drawn from their results. Tell students that their conclusions will help to answer the class question. Ask the students for the definition of a conclusion. They should respond that a conclusion is a claim supported by data. Ask the students how many changing variables an experiment can have to be able to make a claim/conclusion. They should respond that there can only be one changing variable. Ask them why they can only have one changing variable in order to draw a conclusion. They should say that if there is more than one changing variable they would not be able to tell which one of the multiple changing variables affected the smear.

To help learn how to analyze data we will look at another scientists’ data to see if we can draw a conclusion from their results. Have students turn to page 20 of their notebook and place an example notebook under the document camera. Tell the students to look over the data in this results table. Ask a student to explain the experiment that was conducted. Make sure they identify which variables were changing and which were controls. For example one explanation might be: “In this experiment the paper height was changed. For all of the trials, a black dot was put on the original paper 2 cm above the bottom of the paper. The paper was then put in 2 ml of water, in the original test tube, and allowed to sit for 5 minutes. The changing variable was paper height and the controls were the test tube, time, liquid type, liquid amount, paper type, pen color, pen type, and initial dot height.”

Next, draw a picture on the board of what happened in each trial. Ask the students what the paper height was for each trial. Draw papers in descending order labeled A through D. Then ask the students where the dot was on each paper, and draw a line and dot proportionate to the paper sizes at what would be 2 cm. Next ask the students what the liquid height was for each of the trials. For trial A draw a line at 11 cm, for trial B draw a line at 11 cm, for trial C draw a line at 10 cm, and for trial D draw a line at 5 cm. Finally, ask the students what the smear height for each of the trials is as well as where on the paper the smear should be drawn. Ask the students why the smear height is shorter than the water height. Students

should realize that the smear height starts at the line (2 cm from the bottom of the strip) and the water level starts at the bottom of the strip. For each trial the smear should be drawn from the base of the dot to the top of the liquid height (see example drawing below). Ask students if they see any trends in the liquid and smear heights. Students should say that the liquid heights when the paper height was 10 cm and 5 cm is at the top of the paper and the liquid heights when the paper height was 15 cm and 20 cm is at 11 cm. Next have students make several predictions about what would happen to the liquid and smear height if the paper height was several different sizes. For example: if the paper height was 8 cm, would the liquid reach the top of the paper? Yes. If the paper height was 13 cm, would the liquid reach the top of the paper? No. After making several predictions, ask the students what the tallest paper height is that the liquid will reach the top of the paper. This paper height will be 11 cm. After students have a clear idea of what happened in the experiment, ask if a claim/conclusion can be made from this data. Students should realize that there is only one changing variable so a claim/conclusion can be made from these results.



Now that they know a conclusion can be made from the data, you are going to work together to come up with a conclusion. Explain that when drawing a conclusion from data the first step is making a claim to explain the results.

Then ask, “can anyone look at the drawings and make a claim that tries to explain these results?” Below are examples of claims that state how the smear is affected by the paper type.

1. The liquid will reach the top of the paper if the paper is 11 cm or shorter
2. if the paper height is greater than 11 cm the smear height will be 9 cm
3. if the liquid level is the same as the paper height then the smear height will be 2 cm shorter than the paper height
4. the colors that the black dot separates into are the same regardless of the paper height

Below are example of claims that state what happened to the smear.

1. The height of the paper affects the height of the smear
2. The height of the paper does not affect the color of the smear

If possible try to lead the students to a claim that explains how the smear/liquid level was affected instead of just what happened to the smear/liquid level. The underlined claim is the claim that most classes come up with. Tell students that claims that allow you to make predictions are more valuable in science because we can then go out and further test our claims to see if they are correct. Therefore, when they try to generate a claim about their data they should try to have a claim that would allow them to make a prediction. Write the claim in the example notebook and have students copy it into theirs.

Ask the students what data was collected to support this claim.

Below are examples of data to back up claims that stated how the smear was affected.

1. when the paper height was 5 cm and 10 cm the water level was 5 cm and 10 cm (same as the paper height) and when the paper height was 15 cm and 20 cm the liquid level was only 11 cm
2. for trials A (paper height 20 cm) and B (paper height 15 cm), where the liquid height was shorter than the paper, both smear heights were 9 cm
3. for trials C (paper height 10 cm) and D (paper height 5 cm), where the liquid height was the same as the paper height, the smear heights were 8 cm and 3 cm respectively, both were 2 cm shorter than the paper height
4. the color of the smears were observed to be blue, orange and red for all four trials (notice that in order for this statement to be considered data it needed to contain the word observed)

Ask students how we know that the statement we generated is data. Depending on the conclusion drawn they will either say the statement contained a measurement or an observation. Either way, scientists had to go and physically carry out an experiment to discover the results. Write the data in the example notebook and have students copy it into their notebooks. Make sure students understand that the conclusion that they made is the outcome of their experiment and should answer what they were hoping to learn. Tell students that the conclusion of an experiment aims to answer their original question, or answer the class question.

Today students will run their second experiment, complete their bar graph, and draw a conclusion from their data. Remind the students that the group that generates the best claim supported by data or conclusion will receive a prize on the last day. Remind students that they must have their procedure and results table completed before they can start their experiment. Tell students to return to get together with their group and start working.

SCIENTIFIC PRACTICES
Conclusions

Variables:	Trial A	Trial B	Trial C	Trial D
Test Tube:	Original			
Time:	5 min			
Liquid Type:	Water			
Liquid Amount:	2 ml			
Paper Type:	Original			
Pen Color:	Black			
Pen Type:	Mr. Sketch			
Initial Dot Height:	2 cm			
Paper Height:	20 cm	15 cm	10 cm	5 cm
Data:	Trial A	Trial B	Trial C	Trial D
Liquid Height:	11 cm	11 cm	10 cm	5 cm
Smear Height:	9 cm	9 cm	8 cm	3 cm
Other:	blue, orange, red	blue, orange, red	blue, orange, red	blue, orange, red

Write a conclusion from the results above:

We can conclude the liquid will reach the top of the paper if the paper is 11 cm or shorter

because when the paper height was 5 cm and 10 cm the liquid level was 5 cm and 10 cm and when the paper height was 15 cm and 20 cm the liquid level was only 11 cm.

20

RESULTS
Table

Fill out the chart for each of your trials. If one of the variables remains constant for all trials write the value in trial E and then draw a line through each box indicating that this variable is control.

Variables:	Trial E	Trial F	Trial G	Trial H	
Test Tube:	Original				
Time:	5 min				
Liquid Type:	RA	water	soap	vinegar	
Liquid Amount:	3 mL				
Paper Type:	original				
Pen Color:	black				
Pen Type:	crayola				
Initial Dot Height:	2 cm				
Predictions:					
	Trial E	Trial F	Trial G	Trial H	
Put a "T" in the trial that will give the tallest smear and an "S" in the trial that will give the shortest smear.		T	S		
Data:					
	Trial E	Trial F	Trial G	Trial H	
Final Observations/Measurements	Liquid Height:	6.5 cm	8 cm	3 cm	5 cm
	Smear Height:	4.5 cm	6 cm	1 cm	3 cm
	Other:	blue, green, yellow	yellow, blue, green	yellow, blue, green	yellow, blue, green

19

Results Table (5 minutes – Small Groups – SciTrek Volunteers):

If your group has not finished their procedure make sure to do this before moving on to their results table.

Have groups fill in their results table. Make sure that for their controls they are only writing the value of the control in trial E and then drawing a line through the remaining trials. For the changing variable(s) they need to write the value of the variable in each of the boxes. When students have finished, have them make predictions about the final height of the smear. Try to question each group on their thought process behind their predicted order. See example notebook above.

Experiment (20 minutes – Small Groups – SciTrek Volunteers):

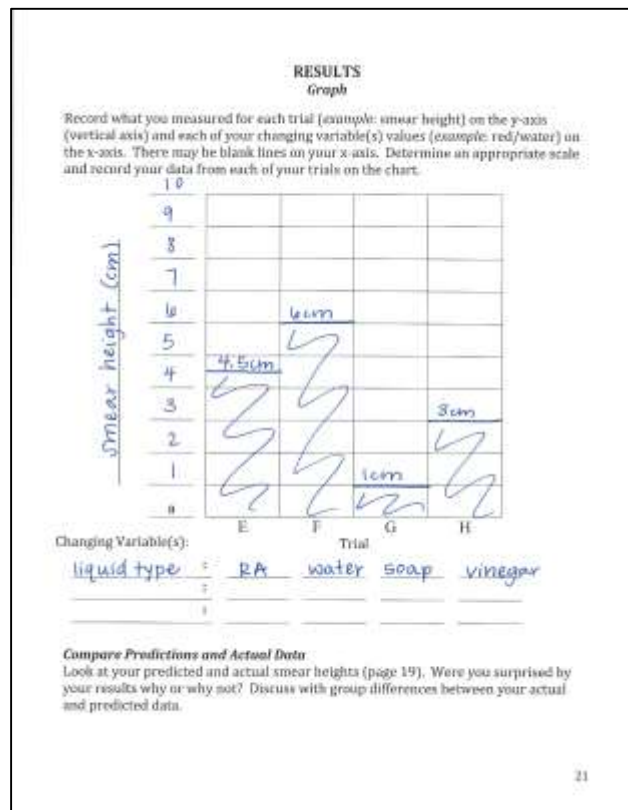
Once groups have completed their results table give them their requested materials. All student materials should be in your group box and all extra materials will be in the lead box. Make sure that students label their strips E, F, G, and H (in pencil at the top of each strip) so they can tell them apart later. As soon as students are done with their liquids, remove the liquids, graduated cylinders, and droppers and put in the bucket (please do not put trash in the bucket). It is important to do this as soon as possible so students do not play with or spill their liquids. When the experiment is finished, place all test tubes in the bucket and put the corks, test tube rack, timers, and pens in your group box. If your group has things under control please help other groups. Once students have finished their experiments, they can record their findings. Make sure that groups trace the liquid line (with pencil) onto their strips so they can easily see/measure it later if needed. Once students have finished their measurements, place their chromatography strips in the Ziploc bag with their group number on it (this is the same bag that their

supplies came in). These chromatography strips will be attached to their posters during the poster making session. Once a group has finished they can move onto graphing their results.

Bar Graph (5 minutes – Small Groups – SciTrek Volunteers):

Help students fill out their bar graphs (page 21). If students pick systems in which the dot did not smear they can go back to page 16 and revise their question to, “What will happen to the height of the liquid?” from, “What will happen to the height of the smear?” They will then be able to plot liquid height instead of smear heights. Once they have graphed their values, make sure that they write the number on top of their line so that it is easy to discern the value.

Once students have completed their bar graph have them use their results table to compare their predicted and actual smear heights and discuss differences between your actual and predicted data. Once they have complete their discussion students can move onto determining their conclusions.



Conclusion (8 minutes – Small Groups – SciTrek Volunteers):

Help the students write their conclusion. First, have them copy their question from page 16 of their notebook onto the conclusions page (page 22). (If they had to change the measurement in their question from smear height to liquid height or vice versa, they can revise their question and record this onto page 16). Second, have the students discuss if their changing variable affected smear/liquid level and how. Challenge students to think about how their changing variable did or did not affect their measurements. For this experiment we will not focus on why the dot is spreading up the paper. Once they have discussed

their ideas have the students fill out the first part on the conclusions page (page 22), “explain how your changing variable did or did not affect your results.”

Example claims that state how the changing variable did or did not affect measurements.

Claim 1: the more absorbent the paper the larger the smear

Claim 2: the more time the taller the smear

Example claims that state what happened to the smear.

Claim 3: the paper type affected the height of the smear

Claim 4: the color of the pen did not affect the height of the smear

If possible have students determine how their changing variable did or did not affect results (claims 1 and 2). If they cannot determine how, have them state what happened to the smear (claims 3 and 4). The best claims will allow students to make predictions about systems that they have not experimented with. Have students verify that their proposed statement is testable. If it is testable then the statement is a claim but if it is not testable then they need to revise their statement.

If there is time, students can determine the data to support their claims. For an example of how to do this, see the conclusion section on Day 6.

Example student work for the conclusion section can be seen below.

What was your question? (page 16)

• If we change the liquid type
insert changing variable
 what will happen to the height and color of the smear?
insert what you are measuring/observing

Explain how your changing variable did or did not affect your results. (Example: the shorter the paper the shorter the smear height.) The thicker the liquid type the smaller the height of the smear

_____ this is your claim

Can you test the above statement? (Is the above statement a claim?)

YES NO (If you checked this box go back and revise your statement so that it can be tested which will make it a claim.)

What data do you have to support your claim? (Remember to include your measurements or observations.) the thick liquid, soap, had a smear height of 1cm while the thin liquid, water, had a smear height of 6cm.

_____ this is your data

How do you know the statement above is data? I know this is data because it contains a measurement.

22

Wrap-Up (2 minutes – Small Groups – SciTrek Lead):

Tell the students that during SciTrek next visit they will get time to finish their conclusion and then creating a poster to share their results with the class.

Clean-Up:

Before you leave, have students attach their nametags to their notebooks and place them in the group box. Collect each groups bag of their experimental strips and put them into the group tub. Make sure that all of the dirty dishes and used liquids are in the bucket and the lid is securely fastened. Bring all materials back to UCSB. In addition, put your lab coat back in your group box.

Day 6: Poster Making

Schedule:

- Introduction (SciTrek Lead) – 2 minutes
- Conclusion (SciTrek Volunteers) – 10 minutes
- Verify Classmates’ Conclusion (SciTrek Volunteers) – 10 minutes
- Poster Making (SciTrek Volunteers) – 35 minutes
- Wrap-Up (SciTrek Lead) – 3 minutes

Materials:

(3) Volunteer Boxes:

- | | | |
|---|---|--|
| <input type="checkbox"/> Student nametags | <input type="checkbox"/> Poster diagram (full page) | <input type="checkbox"/> (9) Paperclips |
| <input type="checkbox"/> Student notebooks | <input type="checkbox"/> (3) Stickers on how to present graph | <input type="checkbox"/> Highlighter |
| <input type="checkbox"/> Volunteer instructions | <input type="checkbox"/> (2) Pencils | <input type="checkbox"/> Scissors |
| <input type="checkbox"/> Volunteer lab coat | | <input type="checkbox"/> (2) Glues |
| (3) Poster Parts Packs | | |
| <input type="checkbox"/> Scientists’ names | <input type="checkbox"/> Procedures | <input type="checkbox"/> Conclusions |
| <input type="checkbox"/> Questions | <input type="checkbox"/> Results | <input type="checkbox"/> (4) “I acted like a scientist when _____” |
| <input type="checkbox"/> Experimental set-ups | <input type="checkbox"/> Results graphs | <input type="checkbox"/> (4) Picture spaces |

Other Supplies:

- Large poster paper

Lead Box:

- | | | |
|--|---|---|
| <input type="checkbox"/> (3) Extra notebooks | <input type="checkbox"/> Poster diagram (full page) | <input type="checkbox"/> (9) Paperclips |
| <input type="checkbox"/> Lead instructions | <input type="checkbox"/> (3) Stickers on how to present graph | <input type="checkbox"/> (2) Highlighters |
| <input type="checkbox"/> Time Card | <input type="checkbox"/> (2) Pencils | <input type="checkbox"/> (2) Scissors |
| <input type="checkbox"/> Lead lab coat | | <input type="checkbox"/> (2) Glues |
| | | <input type="checkbox"/> Scotch tape |

Set-Up:

SciTrek Lead:

Ask the classroom teacher for a place to leave the student posters in the classroom.

SciTrek Volunteer:

Set out SciTrek notebooks/nametags around the table making sure to set students that are in the same small group next to each other. Have poster parts ready for students.

SciTrek Notebook Pages Used with Students:

<p>What was your question? (page 16)</p> <ul style="list-style-type: none"> If we change <u>the liquid type</u> <small>independent variable</small> what will happen to <u>the height and color of the smear</u> <small>(insert what you are measuring/observing)</small> <p>Explain how your changing variable did or did not affect your results. (Example: the shorter the paper the shorter the smear height.) <u>The thicker the liquid type the smaller the height of the smear</u></p> <p style="text-align: center;"><small>this is your claim</small></p> <p>Can you test the above statement? (Is the above statement a claim?)</p> <p><input checked="" type="checkbox"/> YES <input type="checkbox"/> NO <small>(If you checked this box go back and revise your statement so that it can be tested which will make it a claim.)</small></p> <p>What data do you have to support your claim? (Remember to include your measurements or observations.) <u>the thick liquid, soap, had a smear height of 1cm while the thin liquid, water, had a smear height of 4cm.</u></p> <p style="text-align: center;"><small>this is your data</small></p> <p>How do you know the statement above is data? <u>I know this is data because it contains a measurement.</u></p> <p style="text-align: right;">22</p>	<p>Combine your claim and data to get your conclusion.</p> <p style="text-align: center;">CONCLUSION</p> <p>We can conclude <u>the thicker the liquid type</u> <small>claim</small> <u>the smaller the height of the smear.</u></p> <p>because <u>the thick liquid (soap) had a smear</u> <small>data</small> <u>height of 1 cm, while the thin liquid (water) had a smear height of 4cm.</u></p> <p>I acted like a scientist when <u>I wrote a set-up of my experiment and the procedure.</u></p> <p style="text-align: right;">23</p>
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Have a collaborator (classmate) check your conclusion.
Give your notebook to a collaborator and use their notebook to check their conclusion.

The following section will be filled out by your collaborator.

1. Read conclusion (page 23)
2. Identify claim and data. I know the data is the second part of their conclusion because their results contained a measurement
3. Check if the data (in conclusion) matches the results table. Is the data in the conclusion the same as the results table? Yes No
4. Identify number of changing variables. (page 19) The number of changing variables by this group was 1. The changing variable for this group was liquid type
5. Check if data supports the claim. Does the data support the claim? Yes No

Comments: Do you think that tar would cause the liquid height to be even smaller?

Collaborating Scientist's Name: Breanna Johnson

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Introduction (2 minutes – Full Class – SciTrek Lead):

Ask the class, “What is the class question that we have been investigating?” Students should tell you, “What variables affects the smear?” Inform the class that they will be making posters to present their findings to the class. Before they make posters they will have to finish their conclusions and then have a collaborator, a scientist/student studying a similar subject, check their conclusion. At the same time they will check their collaborators conclusion. Ask the class how scientists’ define conclusion, a claim supported by data. Ask the class what a claim is, the explanation of your results, something that can be tested, as well as what data is, measurements or observations. Inform students that when they read their collaborators conclusion they should make sure that the first statement is a claim and can be tested and the second statement, after the “because,” should be data and contain either a measurement or an observation. Once they have checked and committed on their collaborators conclusion they will switch notebooks back. They should then read over the comments and improve their conclusion if needed.

After they have revised their conclusion they will create a poster to present to the class during the next SciTrek visit. This presentation will tell the class what their group has discovered about the class questions. Tell students that they should write as neatly as possible on the poster parts so that the other class members can read their poster. In addition, they will pick one of their two experiments to present. Remind them that they should pick an experiment from which they were able to draw a conclusion.

Conclusion (10 minutes – Full Class – SciTrek Lead):

If students have not made a claim about their data, have them make one. For an example of how this is done, see the conclusion section in Day 5. After students have determined their claim have them

determine the supporting data. This is when students should look at their results table and record in words what measurements or observations were used to support their claim.

Example data to support the four claims that are previously listed.

Data 1: the smear on the paper towel (most absorbent) was 3 cm and the smear on the copy paper (least absorbent) was 0.5 cm

Data 2: the strip that sat in the liquid for 1 minute had a 1 cm smear and the strip that sat in the liquid for 5 minutes had a 4 cm smear

Data 3: the smear on the paper towel was 3 cm and the smear on the copy paper was 0.5 cm

Data 4: all of the smears by the Mr. Sketch pens, regardless of color, were about 3 cm long

Once students have determined the data needed to support their claim, have them justify how they know the statement should be classified as data. (Statements that are data contain a measurement or an observation. If the statement has an observation make sure the word “recorded” or “observed” is in the statement to indicate that the experiment was carried out.) If the data statement does not have a measurement or the words “observed” or “recorded,” have students modify their statement. Then have students combine their claim and data to generate their conclusion.

Example conclusions:

Conclusion 1: The more absorbent the paper the larger the smear because the smear on the paper towel (most absorbent) was 3 cm and the smear on the copy paper (least absorbent) was 0.5 cm.

Conclusion 2: The more time the taller the smear because the strip that sat in the liquid for 1 minute had a 1 cm smear and the strip that sat in the liquid for 5 minutes had a 4 cm smear.

Conclusion 3: The type of paper affected the height of the smear because the smear on the paper towel was 3 cm and the smear on the copy paper was 0.5 cm.

Conclusion 4: The color of the pen did not affect the height of the smear because all of the smears by the Mr. Sketch pens, regardless of color, were about 3 cm long.

Example student work for the conclusion section can be seen below.

What was your question? (page 16)

- If we change the liquid type
independent variable
 what will happen to the height and color of the smear
independent variable

Explain how your changing variable did or did not affect your results. (Example: the shorter the paper the shorter the smear height.) The thicker the liquid type the smaller the height of the smear

this is your claim

Can you test the above statement? (Is the above statement a claim?)

YES NO (If you checked this box go back and revise your statement so that it can be tested which will make it a claim.)

What data do you have to support your claim? (Remember to include your measurements or observations.) the thick liquid, soap, had a smear height of 1cm while the thin liquid, water, had a smear height of 6cm.

this is your data

How do you know the statement above is data? I know this is data because it contains a measurement.

22

Combine your claim and data to get your conclusion.

CONCLUSION

We can conclude the thicker the liquid type
data
the smaller the height of the smear.

because the thick liquid (soap) had a smear
data
height of 1 cm, while the thin liquid
(water) had a smear height of 6cm.

I acted like a scientist when I wrote a set-up of my experiment and the procedure.

23

Select one student to share their conclusion with the class during the wrap-up discussion.

Once students have filled out their conclusion have them fill in the sentence frame (page 23): "I acted like a scientist when _____." This sentence frame should be unique for each of the students.

Verify Classmates' Conclusion (10 minutes – Small Groups – SciTrek Volunteers):

NOTE: If it is later than 15 minutes since the start of the class do not have students complete this section. Skip down to poster making. If there is time after they complete their posters they can go back and complete this section.

Have students collaborate with a student from another group. Have the two students switch notebooks and check each other's conclusions (page 23). When they are done have them return the notebook to the original owner. An example of a checked conclusion is seen below.

Have a collaborator (classmate) check your conclusion.
Give your notebook to a collaborator and use their notebook to check their conclusion.

The following section will be filled out by your collaborator.

1. Read conclusion (page 23)
2. Identify claim and data. I know the data is the second part of their conclusion because their results contained a measurement
3. Check if the data (in conclusion) matches the results table. Is the data in the conclusion the same as the results table? Yes No
4. Identify number of changing variables. (page 19) The number of changing variables by this group was 1. The changing variable for this group was liquid type
5. Check if data supports the claim. Does the data support the claim? Yes No

Comments: Do you think that tar would cause the liquid height to be even smaller?

Collaborating Scientist's Name: Breanna Johnson

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Poster Making (35 minutes – Small Groups – SciTrek Volunteers):

Each small group (three/four students) will make one poster on one of their experiments from which they can draw a conclusion. Assign each student a part of the poster to complete. If a student struggles with writing try to give them a shorter writing section to complete (example: experimental set-up). Have the students write their name on the sections that they complete. In the students’ notebook highlight the section that they will complete/read. If students are completing multiple sections, use the paperclips in your group box to clip together the sections that they are completing/reading this way when presenting they will not need to flip back and forth between pages. Place the following sentence frame sticker on the top of the book of the student that is completing the bar graph (page 21).

The _____
(you should fill this blank out for the student)
 what was measured for _____
 variable was _____
 measurement

Then practice reading the four sentences with that student. For the poster below, the sentence frame would be: The **length of the smear** for **rubbing alcohol** was **4.5 cm**. Make sure that you fill in the “length of the smear” for the student in the sentence frame but leave the “variable” and “measurement” blanks empty.

When students finish writing their sections, have them draw a picture of their experiment or how they acted like a scientist.

As soon as students have completed some of their pieces start gluing them onto the large poster paper exactly as they are arranged in the example below. Do not wait until students have completed all the pieces to start gluing them onto the poster.

Once the poster is complete, have students start practicing for the presentation. Make sure that students read from their notebook instead of the poster. The poster should be presented in the following order: 1) scientists' names, 2) question, 3) experimental set-up, 4) procedure, 5) results graph, and 6) conclusion. They will NOT read the "I acted like a scientist when _____" or results table sections from their poster. The "I acted like a scientist when _____" section will be discussed as a class after all posters are presented. In addition, it sometimes helps to write a small number (order seen above) on the sections in the students' notebooks in the order that they will be presented.

The image shows a collage of various sections from student posters, arranged as they would appear on a large poster. The sections include:

- Question:** A student's question about liquid height and smear length.
- Experimental Set-up:** A diagram showing four test tubes labeled A, B, C, and D, with a list of materials: liquid (soap, water, honey), paper (soap, paper), and a ruler.
- Procedure:** A list of steps: 1. Pour liquid into test tubes, 2. Dip ruler, 3. Smear, 4. Measure smear and record length.
- Results:** A table with columns for Smear, Soap, Water, Honey, and Smear Height. The data shows that soap has the smallest smear (1 cm) and the tallest smear height (6 cm), while water has the largest smear (6 cm) and the shortest smear height (1 cm).
- Conclusion:** A student's conclusion stating that thicker liquids result in smaller smears.
- Scientists' Names:** A section listing the names of Student 1, Student 2, and Student 3.
- Picture:** A box labeled "Picture" for each student, intended for a drawing related to the experiment.

If there is time try to ask each of your groups a few questions about their poster. Have them use their findings to predict what would happened to the smear for other experiments that they did not perform but are related to their experiment. For instance if the group conclusion was, "the thicker the liquid the smaller the smear because soap gave a 1 cm smear height and water gave a 6 cm smear height," ask the group to predict what length would the smear be if your liquid was honey. They should be able to predict that it would be less than 1 cm because honey is thicker than soap.

Wrap-Up (3 minutes – Full Class – SciTrek Lead):

Go around and have one student from each of the groups share their conclusion with the rest of the class. Once a conclusion is shared have students verify that the claim and data statements are correct. Tell students that they will present their findings during the next SciTrek visit and that you are looking forward to hearing about their experiments.

Clean-Up:

Before you leave, have students attach their nametags to their notebooks and place them in the group box. Leave student posters in the classroom. Bring all materials back to UCSB. Look through all of the conclusions and place a notebook from the group with the best conclusion on the top of your box. In addition, put your lab coat back in your group box.

Day 7: Poster Presentations**Schedule:**

Introduction (SciTrek Lead) – 2 minutes
Practice Posters (SciTrek Volunteers) – 5 minutes
Poster Presentations (SciTrek Volunteers/SciTrek Lead) – 48 minutes
Wrap-Up (SciTrek Lead) – 5 minutes

Materials:

(3) Volunteer Boxes:

- | | | |
|---|---|--|
| <input type="checkbox"/> Student nametags | <input type="checkbox"/> Volunteer lab coat | <input type="checkbox"/> Large binder clip |
| <input type="checkbox"/> Student notebooks | <input type="checkbox"/> Pencil | <input type="checkbox"/> Highlighter |
| <input type="checkbox"/> Volunteer instructions | <input type="checkbox"/> (9) Paperclips | <input type="checkbox"/> (12) Sharpened SciTrek pencils (all same color) |

Lead Box:

- | | | |
|--|---|--|
| <input type="checkbox"/> (3) Extra student notebooks | <input type="checkbox"/> Lead lab coat | <input type="checkbox"/> (9) Paperclips |
| <input type="checkbox"/> Lead instructions | <input type="checkbox"/> (9) Stickers on how to present graph | <input type="checkbox"/> (2) Highlighters |
| <input type="checkbox"/> Time card | <input type="checkbox"/> (2) Pencils | <input type="checkbox"/> Scotch tape |
| | | <input type="checkbox"/> (4) SciTrek erasers |

Student posters should already be in the classroom.

Set-Up:

SciTrek Lead:

Write the class question on the board, “What variables affect the smear?” Leave enough room to record student findings under the question.

SciTrek Volunteer:

Set out the SciTrek notebooks/nametags. Get your groups' posters. Today students will be sitting in their regular classroom seats during poster presentation. Have pencils ready to distribute to your group after the poster presentations.

Introduction (2 minutes – Full Class – SciTrek Lead):

Tell students that today they will present their posters to the class. Inform students that this is a common practice in science. Scientists go to conferences where they present posters. At these presentations other scientists give them feedback on their experiments which allows them to return to the lab with new ideas for future experiments.

Tell the students that they will have 5 minutes to discuss their experiment and results and practice presenting their poster with their group. While discussing their experiments and results students should not look at their notebooks or poster. Remind students to read from their notebooks when presenting. Tell students that after practicing they will return to their normal classroom seats.

Practice Posters (5 minutes – Small Groups – SciTrek Volunteers):

Make sure that each small group is telling each other what they did and what they learned from their experiment. Then have them practice their poster presentation, making sure they are reading the poster parts in the correct order (scientists' names, question, experimental set-up, procedure, results graph, and conclusion). Make sure each student section is highlighted in their notebook. If students are reading from multiple pages use a paperclip to clip these pages together. Remind students to read from their notebook rather than from their poster.

Poster Presentations (48 minutes – Full Class – SciTrek Volunteers/SciTrek Lead):

Have students return to their original class seats. Ask the class, "What question are we working on solving?" Students should tell you, "What variables affect the smear?" Tell the students, after each presentation you will ask the class what changing variable the group was testing. They will then be given the opportunity to ask scientific questions to the presenting group to help determine if the variable investigated affected the smear. Students that ask a scientific question during the presentation will get a SciTrek pencil at the end of the presentations. Give the class one or two examples of scientific questions. Some possible examples are: Why did your group pick the controls that you did? Can you make a prediction about another paper type that you think would give a large smear height? Did your results agree with a previous group's and why? Ask the class, "If a group's changing variable was liquid type, would they be able to answer questions about changing the paper type?" They should say no. "Therefore, we should try to ask the group questions that focus on their changing variable." Tell the class once they have finished asking the group questions you will ask them if the groups changing variable affected the height of the smear and how. You will then summarize their ideas on the board under the class question.

Volunteers should make sure that students are quiet and respectful when other groups are presenting. When one of your groups is presenting go to the front of the room with them and prompt students if they get stuck and remind them to read from their notebooks.

During the student question time, the SciTrek lead and/or volunteers should ask at least one question. Example of possible question are, "How do you know...?" or "Is there anything else you can do to get more information about your question?" Each group should answer approximately four questions (one questions per student).

After all poster presentations have been given, ask the class, "What did we learn about the smear?" Have them summarize the class findings. The highlights from many experiments are seen below. Do not expect students to know highlights from experiments that were not run.

- The thinner the liquid the taller the liquid/smear height.
- Different liquids will change the order the colors appear in the smear, but not what colors appear.
- The more absorbent the paper the taller the liquid/smear height.
- If the liquid level is above the dot, the smear will travel downwards. If the liquid level is below the dot, the smear will travel upwards.
- Different black pens are made up of different dye colors.
- Colored markers separate into fewer colors than black markers.
- The longer the time the taller the liquid/smear height.

While summarizing experiments, use students' collected data and not what they should have found from the list above. Tell students that you want to get the longest most colorful smear and that you need them to tell you what values of variables you should use.

- Time: As long as possible
- Liquid type: Water or another thin liquid
- Liquid Amount: Enough to get the liquid close to the dot without going over the dot
- Paper Type: The original paper or another absorbent paper
- Pen Color: Black
- Pen Type: Any washable marker
- Initial Dot Height: A value that would put the dot close to the liquid level

If no one in the class did experiments on one of the variables above, then they will not know how that variable affects the pen dye and do not expect them to tell you which value to use. Tell students that they have taught you a lot about pens and their ink.

Each SciTrek volunteer should keep track of who asks a question and give these students a pencil at the end of the session. If a student does not ask a question during the poster presentations have them ask/answer a question about the experiment before you give them a pencil.

Wrap-Up (5 minutes – Full Class – SciTrek Lead):

Ask students the following question:

How did you act like a scientist during this project?

Tell the students that the volunteers that have been working with them are undergraduate and graduate students that volunteer their time so that they can do experiments. Have the students say thank you to the volunteers. This is the last day with their SciTrek volunteers, therefore, they should say goodbye to them. Tell students that you will be back one more time.

Announce the group that came up with the best conclusion, or claim supported data, from their experiments. Talk a little about why that group's conclusion was selected. Distribute prizes (erasers) to that group.

Tell students to remove the paper part of their nametag from the plastic holder. They can keep the paper nametag, but need to give the plastic sleeve back to their SciTrek volunteer.

Clean- Up:

Before you leave, collect the plastic parts of nametags and put them in the group box. Students can keep the paper part of their nametag. Also collect notebooks and clip them together with a large binder clip. Leave student posters in the classroom. Bring all materials back to UCSB. Remove tape from the lid of your box and place into your box. In addition, remove all materials from lab coat pockets, remove your nametag, unroll lab coat sleeves, and put your lab coat back in your group box.

Day 8: Conclusion Assessment/Tie to Standards

Schedule:

Conclusion Assessment (SciTrek Lead) – 10 minutes

Tie to Standards (SciTrek Lead) – 50 minutes

Materials:

Lead Box:

- | | | |
|--|---|---|
| <input type="checkbox"/> (3) Extra student notebooks | <input type="checkbox"/> (2) Pencil | <input type="checkbox"/> (20) Pure substance bags labeled (sugar, salt, baking soda, and corn starch) |
| <input type="checkbox"/> Student notebooks | <input type="checkbox"/> (2) Mystery data sheets | <input type="checkbox"/> (20) Pure substance bags labeled with letters (A=baking soda, B= sugar, C= salt, and D= corn starch) |
| <input type="checkbox"/> Lead instructions | <input type="checkbox"/> Rock showing a visible mixture | |
| <input type="checkbox"/> (35) Conclusion assessments | <input type="checkbox"/> Jar of sand and water | |
| <input type="checkbox"/> Time card | <input type="checkbox"/> (21) Mixture bags (7 bags of each of the 3 different mixtures) | |
| <input type="checkbox"/> Lead lab coat | | |

SciTrek Notebook Pages Used With Students:

TIE TO STANDARDS

1. Circle the value of the variable that the police should use to process the evidence from the suspects that would give them the longest smear.

Time: 3 min 5 min 10 min All would give similar height smears

Liquid Type: Water Soap Syrup All would give similar height smears

Paper Type: Paper Towel Sandpaper Cardboard All would give similar height smears

Dot Size: All would give similar height smears

Amount of Liquid/Line Level: All would give similar height smears

2. What conclusion can you make from the results the police collected?

We can conclude that the robber was number four data because it was observed that ink from the letter and number four were the same height. data

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3. List one substance that you can see is a mixture, and identify the individual parts. Example: Mixture: pencil, Parts: graphite and wood.
Mixture: lucky charms Parts: cereal & marshmallows

4. List one substance that you can NOT see that is a mixture and identify the individual parts. Example: Mixture: air, Parts: nitrogen, oxygen, and carbon dioxide.
Mixture: soda Parts: liquid and CO₂ gas

5. Take each mixture and separate them into two parts. Be prepared to explain what properties you used to separate each mixture. There can be more than one property that you used.

Number:	Mixture:	What did you use to separate your mixture? <i>(physical property)</i>
1		state of matter, color
2		texture, size
3		weight, material, color
4		magnetism, thickness, color

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6. Write down physical properties of the four substances. You will use these to identify four unknown substances. You will not have access to the labeled substances when you are identifying the unknown substances.

Pure Substance	Physical Properties	Unknown Letter
Sugar	brownish, small granules, many different shaped pieces	B
Salt	white, small granules, square shaped pieces	C
Baking Soda	white powder	A
Corn Starch	white powder, sticks to side of bag, crunches when pinched	D

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7. Match the following key terms with the correct definition by drawing a line between each pair.

Key Terms

1. Pure Substance

2. Mixture

3. Physical Property

Definitions

a. A material made up of two or more types of substances.
Examples: Oxygen & Helium Water & Sand

b. A property that can be measured or observed without changing the substance.
Example: Boiling Point (temperature at which a substance boils)

c. A material made up of only one type of substance.
Examples: Helium (He) Carbon Dioxide (CO₂)

Diagram showing connections:
 1. Pure Substance → c. A material made up of only one type of substance.
 2. Mixture → a. A material made up of two or more types of substances.
 3. Physical Property → b. A property that can be measured or observed without changing the substance.

B. Fill in the following diagram

```

    graph TD
      Matter[matter] --> Mixture[mixture]
      Matter --> PureSubstance[pure substance]
      Mixture -- physical property --> PureSubstance
      PureSubstance --> Molecules[molecules or compounds are H2O or NaCl]
      PureSubstance --> Atoms[atoms are C, O, H, and S]
      Molecules -- chemical reactions --> Atoms
    
```

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Set-Up:

SciTrek Lead:

Pass out notebooks to students. If you do not have time to get set-up before the start of the module ask the teacher to pass out notebooks during the conclusion assessment.

If the classroom has a document camera, ask the teacher to use it to fill out the tie to standards activity with students on pages 25-28. If the classroom does not have a document camera, then tape example poster size notebook pages to the front board.

Conclusion Assessment (10 minutes – Full Class – SciTrek Lead):

“Before we start our activity today we will determine how your ideas on conclusions are developing. One of the ways that we get program funding is by demonstrating the program effectiveness. Therefore, we need you to do your best on the assessment.” Pass-out the conclusion assessment and tell students to fill out their name, teacher’s name, and date on the top of the assessment. Remind the students that it is important that they fill out this assessment on their own. For section one, read the instructions to the students. Then read each of the statements and tell the students to circle if the statement is an example of a claim, data, or neither. As you are reading the statements walk around the room and verify that students have written their name on the top of the paper. For section two, read the instructions to the students and give them a few minutes to make matches between claims and data. You do not need to read the individual claim and data statements to the students. For section three, read the instructions to the students. Then read each of the statements. Tell students based on the results table to circle if the statement is an example of claim, data, opinion, or if it is incorrect. You do not need to read or review the results tables with the students. Read the last question to the students and have them fill in the blank. When students are finished, collect the assessments and verify that the student’s name is on the top of the paper.

Tie to Standards (50 minutes – Full Class – SciTrek Lead):**Mysterious Robbery (15 minutes)**

Start off the discussion by giving the students a mystery problem that they can help to solve by using their experimental results. Tell the students, “10 years ago a robbery happened that was never solved. The police have contacted me to solve the cold case. At the time of the crime a note, written in black pen, was passed to the teller which read “Give me all your money or else.” The teller handed over the money but kept the note. In the confusion that followed, the robber managed to get away. At the time there were eight suspects. Each of these suspects was found with a black pen on them (which the police still have). The only other evidence that the police have from the original crime was the note. No fingerprints were found on the note or at the scene of the crime. How could you figure out who wrote this note, using the skills and knowledge learned from your experiments?” If students are having trouble coming up with responses prompt them by asking what they did in their experiments and have students expand on these ideas. Allow students to discover that you can determine the identity of the thief by determining the unique properties of each pen. If you can find a link between the ink on the letter and the ink from one of the pens found on the suspects then you can identify the robber. Just like the experiments they did, they could take a sample of the ink from the letter and use this as the control. They could then take ink from each of the pens (changing variable) that were found on the suspects and run an experiment. The pen

that has matching ink to the control will be the robber of the bank. Make sure students understand that you would be able to identify the robber by using the unique physical properties of each of the black pens.

Ask students if it would be easier to identify the thief's pen from a longer or small smear? The students should realize, the longer the smear the more details they could learn about the dyes used in the pen. This would make it easier to tell the pens apart and identify the thief. Tell the students that the police have suggested several variables that could be used to run the experiment, but the police need the students help in identifying the best values of the variables. Have students turn to page 20 of their notebooks and fill out the chart together.

Ask students which amount of time would allow the smear to be the longest. Students should realize that the longer the time in the liquid, the larger the smear. Thus, the "best" choice for time will be 10 minutes. Circle 10 minutes in the example notebook and have students copy this into their own notebook. An example can be seen below.

1. Circle the value of the variable that the police should use to process the evidence from the suspects that would give them the longest smear.

Time: 3 min 5 min 10 min All would give similar height smears

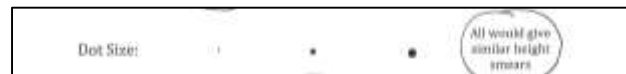
Now ask students which liquid type would allow the smear to be the longest. Students should realize that the thinner the liquid the larger the smear because thick liquids do not travel as far up the paper as the thinner liquids. Student should have experienced this if they tried soap (thick liquid) in comparison to water, vinegar, or rubbing alcohol (thin liquids). Syrup is another thick liquid therefore, students should be able to predict that it will not be a "good" liquid. From the selections given, water would be the best choice for the longest smear. Circle water in the example student notebook and have students copy this response into their own notebooks. An example can be seen below.

Liquid Type: Water Soap Syrup All would give similar height smears

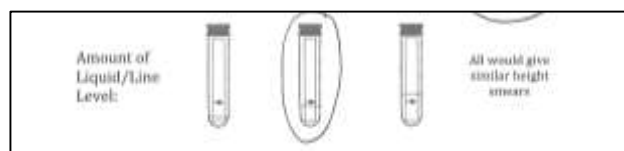
Now ask students which paper type would allow the smear to be the longest. From their experiments students should have seen that the more absorbent a paper is the farther the liquid can travel up the paper. Papers from their experiments that resulted in long smears were paper towel, coffee filter, and the original paper. Paper from their experiments that resulted in short smears were copy paper (any color), graph, construction, and newspaper. The most absorbent paper that they have to choose from is paper towel so this is the "best" choice. Circle this response in the example notebook and have students copy this response into their notebook. An example can be seen below.

Paper Type: Paper Towel Sandpaper Cardboard All would give similar height smears

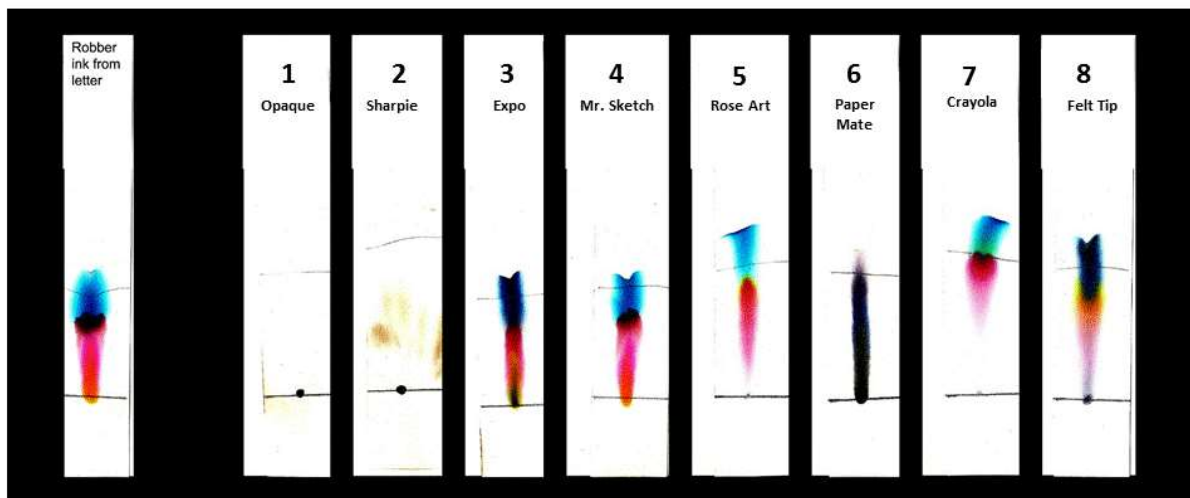
Ask students which dot size would allow the smear to be the longest. Students should realize that the dot size will not affect the height of the smear. However, dot size would have an effect on the width of the smear. Therefore, dot size would give a smear of about the same height. Circle this response in the example notebook and have students copy this response into their notebook. An example can be seen below.



Now ask students how much liquid should be used to get the longest smear. From their experiments students should have seen the lower the liquid level, the longer time it takes for the liquid to reach the dot. This leaves less time for the dot to smear, resulting in a shorter smear height. However, if the liquid level is above the dot, and the dot is soluble in that liquid (smears in that liquid), the dot will start to dissolve into the liquid and will travel down into the liquid instead of up the paper. Therefore, the liquid level should be as close to the dot as possible without going over the dot. Circle this response in the example notebook and have students copy this response into their notebooks. An example can be seen below.



Once students have filled out the table tell them that the police took their suggestions and ran their experiments. Show students the results of the experiment.



Have students look at the data and come up with a conclusion. Because there are no measurements on the strips students will have to use observations for data. Therefore, make sure that they use the word observation in the data portion of the conclusion. Have students individually fill out the conclusions and then share them with the class. An example conclusion is seen below.

2. What conclusion can you make from the results the police collected?

We can conclude that the rubber was number four
down
 because the ink from the letter and number four
were observed to be on the same height.

Mixture Discussion (5 minutes)

What do you think the results imply about composition of the black ink? Students should respond, the black inks are made up of many different colored dyes making them all different mixtures. All materials can either be divided into mixtures or pure substances. Pure substances are made up of one type of atom or molecule and mixtures are made up of two or more distinct pure substances. Some mixtures are easily identified because you can see the different substances. Show students the rock in which the individual components of the mixture can be seen. Have students come up with four examples of mixtures they can see with their eyes and record one in the example notebook. Tell students that they can either record the mixture that you wrote or write another example in their notebook. An example can be seen below.


3. List one substance that you can see is a mixture, and identify the individual parts. Example: Mixture: pencil Parts: graphite and wood
 Mixture: LUCKY CHARMS Parts: sereal and marshmallows

Tell students that sometimes you cannot see the individual parts of a mixture until it is separated, like the ink from the black pens. Now have students come up with four examples of a mixtures they cannot see with their eyes and record one in the example notebook. Tell students that they can either record the mixture that you wrote or write another example in their notebook. An example can be seen below.

4. List one substance that you can NOT see that is a mixture and identify the individual parts. Example: Mixture: air Parts: nitrogen, oxygen, and carbon dioxide
 Mixture: soda Parts: liquid and CO₂ gas

Separating Mixtures/Physical Properties (10 minutes)

Hold up the bottle of water and sand. Ask the students if the bottle contains a mixture or a pure substances and why. They should be able to see that the bottle contains two different substances, therefore, it is a mixture. Ask them what properties might they use to separate the two substances. Students should notice that the water is a liquid and the sand is a solid. If they do not know what this is called tell them that scientist would say that the water and the sand are different states of matter. They also might notice that the two substances vary by other properties including color, texture, opacity, etc. Record their responses into the example notebook and have students copy this response. Record these properties next to mixture one on the example group notepad and have students copy them into their notebooks.

Number:	Mixture:	What did you use to separate your mixture? (physical property)
1		state of matter, color

Tell the students they are now going to get a baggie that contains a mixture and they then need to think about all of the properties of those substances in the bag, that they could use to separate the substance by, and then write these properties down next to the picture.

Pass out the mixture bags, one bag to every two students. There are three different mixtures bags that students could get, try to keep students that are at the same table with the same mixture bag. Give students three minutes to complete the task. At the end of the three minutes have students put the materials back into the bag and collect all of the bags from the students.

Ask students with mixture 2 which properties they used for separating the mixture. Students should have been able to feel that the different sheets of sandpaper have different textures. They also might notice that there is a slight variation in the color of the two types of sandpaper. Record their responses into the example notebook and have students copy this response. An example can be seen below.



Ask students with mixture 3 which properties they used for separating the mixture. Students should notice that the two types of nuts different materials and different colors. In addition, the two nuts have different masses. Record their responses into the example notebook and have students copy this response. An example can be seen below.



Ask students with mixture 4 which properties they used for separating the mixture. Students should notice that the different substances vary in color, and thickness. They should also see that the magnets stick to each other while the papers do not. Record their responses into the example notebook and have students copy this response. An example can be seen below.



Tell students that they just separated the mixtures using the physical properties of the individual substances in the mixture. Have them write “physical property” at the top of chart. Tell students that physical properties are attributes of the material that you can measure or observe without changing the substance. Other physical properties include melting point, boiling point, solubility, conductivity, hardness, etc.

Tell students that physical properties are important because they can be used to separate mixtures. Then give three examples:

Magnetism can be used to separate mixtures. If you had a sand and iron mixture you could put a magnet into the mixture. This would cause all of the iron to stick to the magnet and then you could pour off the sand.

Different states of matter can be used to separate mixtures. If you had a sand and water mixture the mixture could be poured through a filter. The liquid would pass through and the solid would be left behind.

Boiling point can be used to separate mixtures. If you had a water and a salt mixture you could heat the mixture up. This would cause the water to evaporate leaving behind the salt which has a much higher boiling point.

In our experiment, we separated out the different dye colors in a pen. Let's try to determine the physical properties that were used to separate the dyes.

What did we do to get the dye colors to travel up the paper?

We had to put the paper in liquid/water.

If there was red on the top of the smear and green on the bottom of the smear which color was more attracted to the water and why? (If needed draw a picture on the board)

The red traveled farther up the paper so it must be more attracted to the liquid/water.

What happened when Sharpie pens were tried with water?

Nothing /the dot did not move.

Were the Sharpie pens attracted to the water and how do you know?

No because it did not go up the paper.

Do you think that the Sharpie pens are attracted to anything?

The Sharpie marker must have been attracted to the paper because it did not move.

What happened when you put the Sharpie in the rubbing alcohol?

It had a small smear.

Explain to the students that this means that the Sharpie ink was attracted to the rubbing alcohol. Explain them that different dyes are attracted to different liquids, therefore, if you change liquids you change the attraction, which can change the order colors appear in the smear. For example water could yield a smear that is blue, green, yellow and rubbing alcohol could yield a smear that is green, blue, yellow.

Be sure that by the end of the conversation students understand that the physical properties used to separate dye colors were the attraction of the ink to the paper and/or the attraction of the ink to the liquid. Then inform the class that this method of separation is called chromatography.

Pure Substances (15 minutes)

Tell students that physical properties can be used to separate mixtures, but they also can be used to identify pure substances. Pass-out the labeled pure-substance bags; these four bags contain a single pure substance (sugar, salt, baking soda, and powder). Students will be given a few minutes to identify physical properties of each of the pure substance which they should record in their notebooks. Tell them that they may feel and look at the bags, but they may not open any of the small bags. Once they have written down the physical properties, you will take away the bags and give them four bags that are labeled with only a letter. They will then need to determine the identity of the pure substance based off their notes. Give students ~5 minutes to examine and write down the physical properties of the labeled substance, then remove the bags. Hand out the letter bags and have them determine what the unknown substances are:

Bag A = Baking Soda

Bag B = Sugar

Bag C = Salt

Bag D = Corn Starch

Once students have completed the activity ask them to share their prediction for each lettered bag. Also ask them what physical property was most helpful in identifying the substance. Once a student has shared, poll the class using thumbs up/thumbs down for agree/disagree with the student. If a student disagrees have them explain why and share what physical property they used to identify the substance. Collect the bags from the students.





6. Write down physical properties of the four substances. You will use these to identify four unknown substances. You will not have access to the labeled substances when you are identifying the unknown substances.

Pure Substance	Physical Properties	Unknown Letter
Sugar	brownish, small granules, many different shaped pieces	B
Salt	white, small granules, square shaped pieces	C
Baking Soda	white powder	A
Corn Starch	white powder, sticks to side of bag, crunches when pinched	D

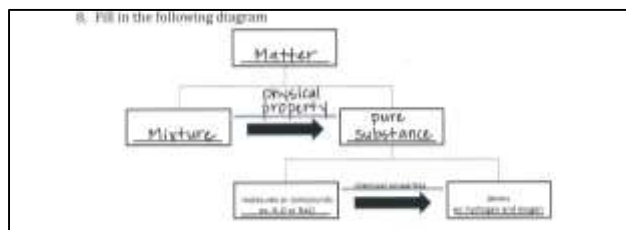
Definition Discussion (5 minutes)

Have the students turn to page 28 in their notebooks and have them work by themselves to match the following terms with their correct definitions. After the students have had a few minutes, review the answers as a class. Draw a line between each definition on the example notebook under the document camera so that students can copy the correct answers. An example notebook can be seen below.

7. Match the following key terms with the correct definition by drawing a line between each pair.

Key Terms	Definitions
1. Pure Substance	a. A material made up of two or more types of substances. Examples: Oxygen & Helium Water & Sand  
2. Mixture	b. A property that can be measured or observed without changing the substance. Example: Boiling Point (temperature at which a substance boils)
3. Physical Property	c. A material made up of only one type of substance. Examples: Helium (He) Carbon Dioxide (CO ₂)  

Ask the students if they know the term to describe materials with mass (matter). Have them write “matter” in the top box of question eight. Explain that matter can be split into two categories, both of which we have talked about today. Ask them if they remember these two categories. Students should reply mixtures and pure substances. Tell the class that mixtures can be separated into pure substances. Ask the class what properties we take advantage of to perform these separations. Students should state physical properties. Make sure students fill in all the terms in question eight. An example can be seen below.



If molecules/compounds have come up in your discussion you can also briefly discuss how chemical properties can be used to separate molecules/compounds into pure elements.

Tell the class that you have enjoyed learning science with them and they will get another opportunity for SciTrek to come to their room and run long term investigations with them later in the year. Tell them to remember what they have learned for their next module.

Clean- Up:

Bring all materials back to UCSB.